

Organocatalysis - A New Tool for Industry

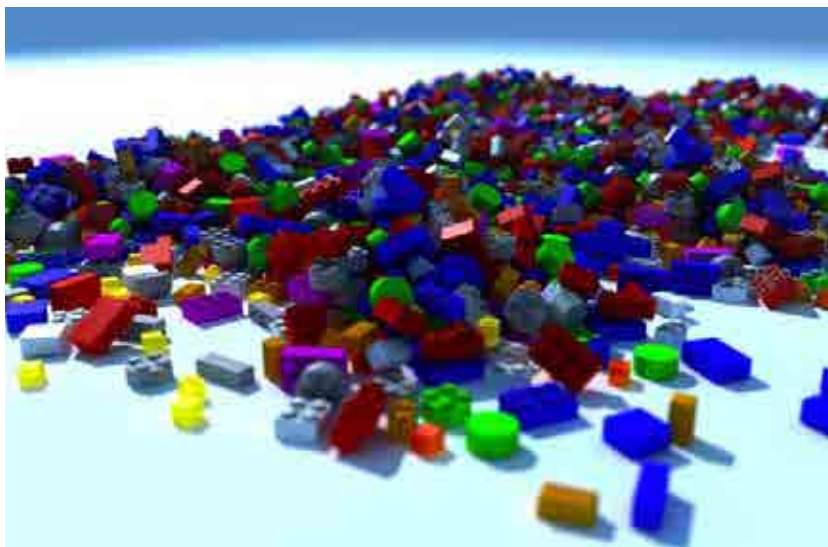


The Royal Society of Chemistry Symposium
Catalysts for Change

What is my mission

What is organocatalysis

To present an overview organocatalytic reactions



To show the synthesis of optically active building blocks and their application

To use simple and easy available starting compounds and catalysts

Organocatalysis

Advantages

- a) Easy to prepare
- b) Easy to handle - water, air
- c) Easy to scale up
- d) No metal contamination
- e) Easy screening
- f) "More flexible" than metal catalysis
- g) More complex reactions and multiple stereocenter formation

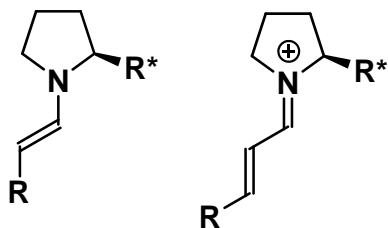
Disadvantages

- a) High catalyst loadings
- b) "Premature"

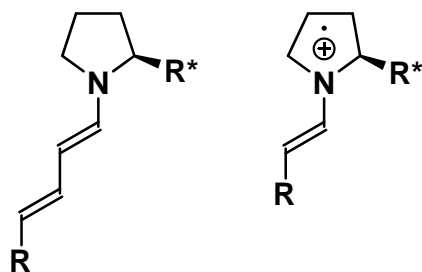
Organocatalysis - activation

Covalent catalysis

enamine iminium

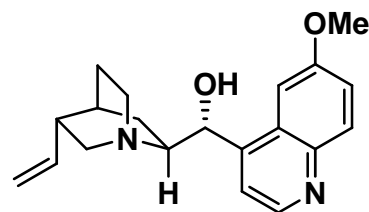


dienamine SOMO



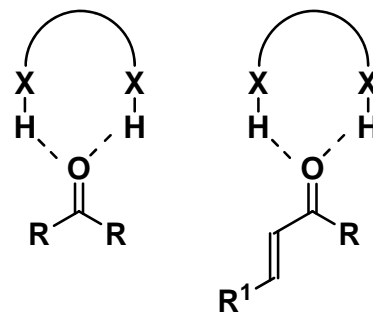
Non-covalent catalysis

cinchona alkaloids/PTC



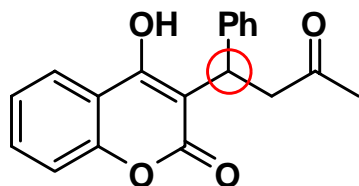
Quinine

proton activation



An "old" example - Warfarin - from lab to humans

100% atom economic, one-step synthesis of optically active anticoagulants



Warfarin
sold as racemate for 40 years

Activity: (*S*)-enantiomer is about 5 times higher than the activity of the (*R*)-enantiomer, and the enantiomers are metabolized by different metabolic pathways.

Different half-lives of the enantiomers: 21-43 and 37-89 h for (*S*)- and (*R*)-warfarin, respectively.

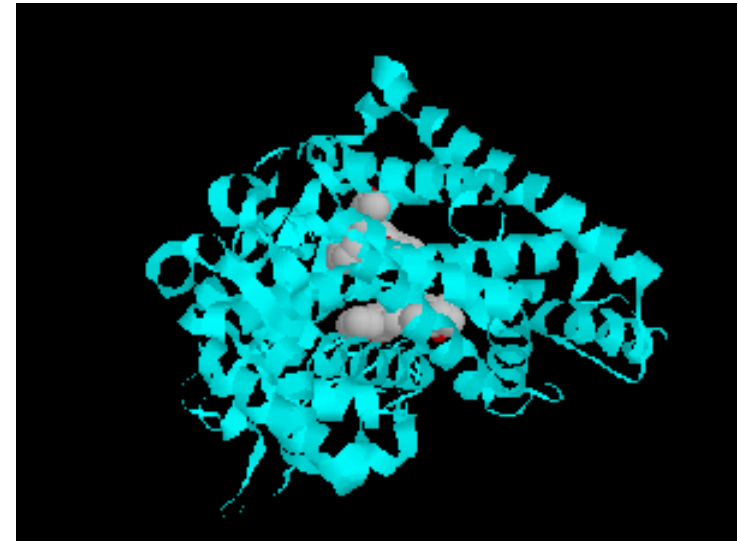
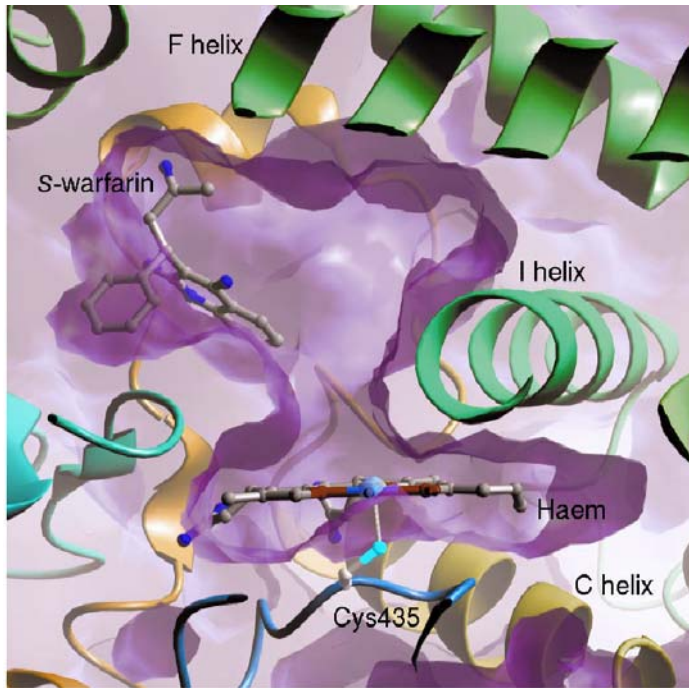
Problem: patients are responding very differently to racemic warfarin.

Treatment of patients with optically pure warfarin will reduce the dose problem.

Advantage: weak patients, unable to tolerate the stronger racemic or (*S*)-warfarin, could be treated with the milder (*R*)-warfarin.

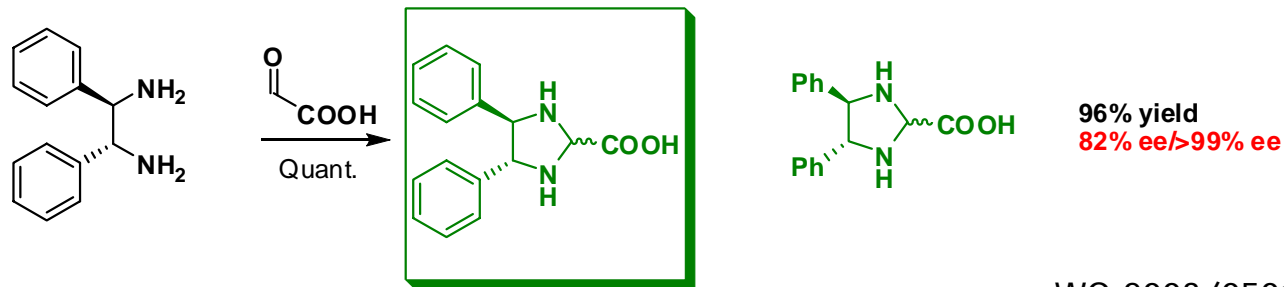
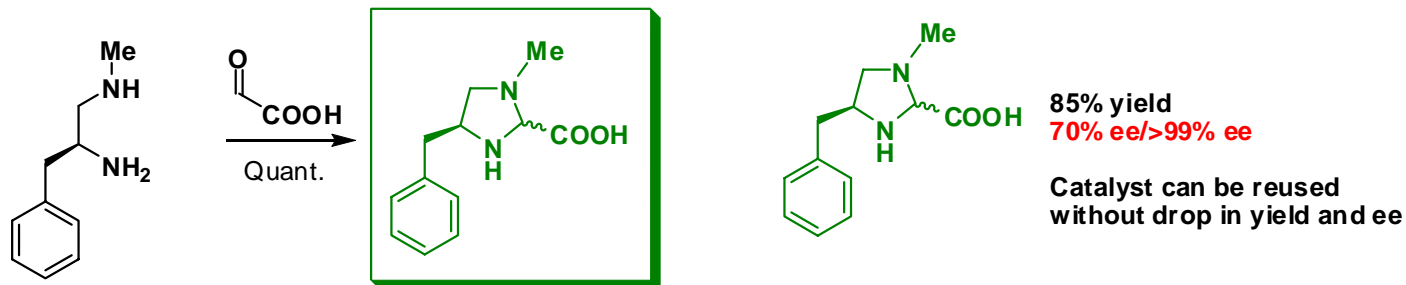
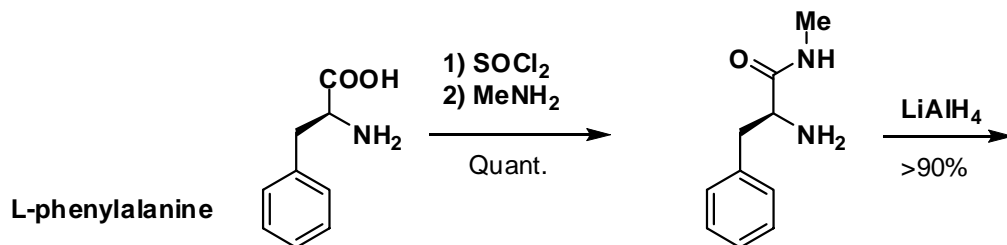
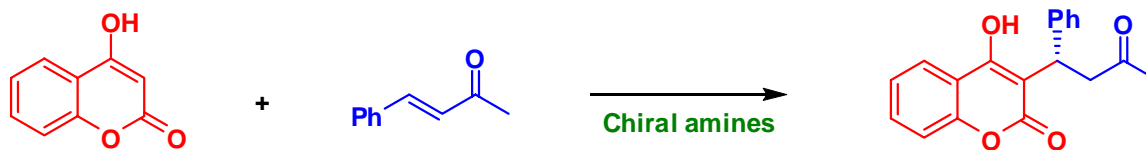
The most important advantage of administering optically pure warfarin is the possibility of eliminating drug-drug interactions, which represents another serious problem with racemic warfarin, as the cytochrome P-450 enzyme responsible for metabolising warfarin is also metabolizing other drugs

S-Warfarin metabolized by cytochrome P450

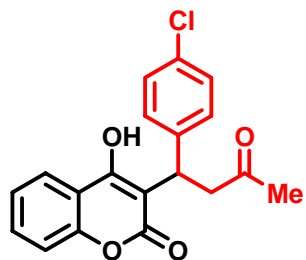


P. A Willams et al. *Nature*, 2003, 424, 464

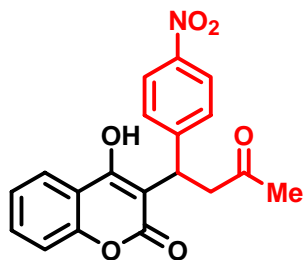
Development of catalysts



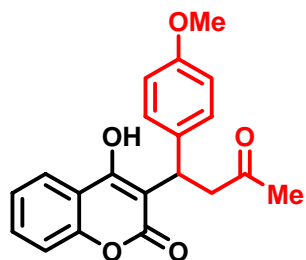
Variation of enones and coumarins



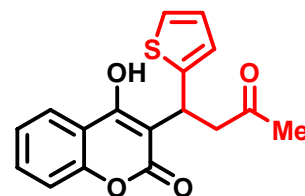
68% yield
87% ee



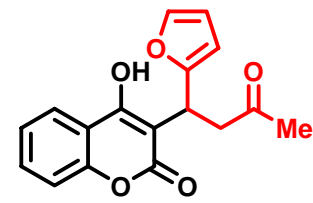
81% yield
83% ee



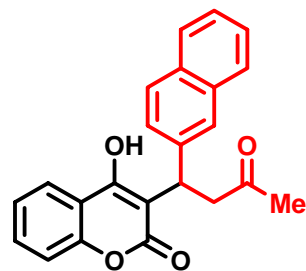
91% yield
82% ee



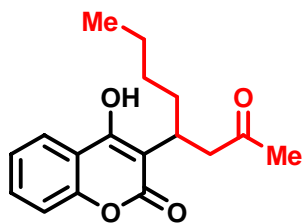
78% yield
79% ee



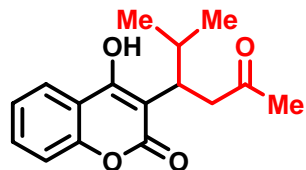
75% yield
76% ee



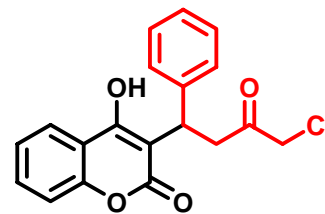
81% yield
83% ee



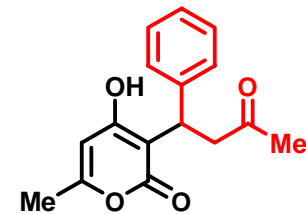
99% yield
84% ee



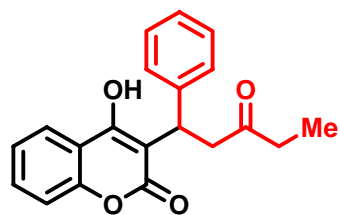
77% yield
83% ee



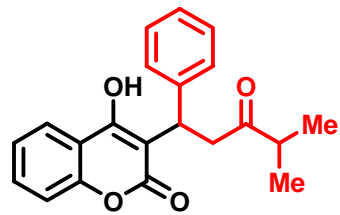
12% yield
84% ee



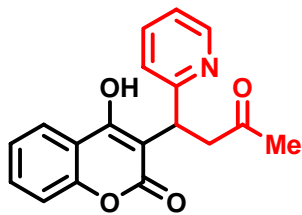
76% yield
85% ee



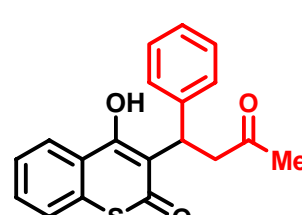
84% yield
82% ee



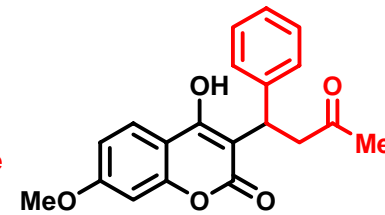
71% yield
85% ee



68% yield
87% ee



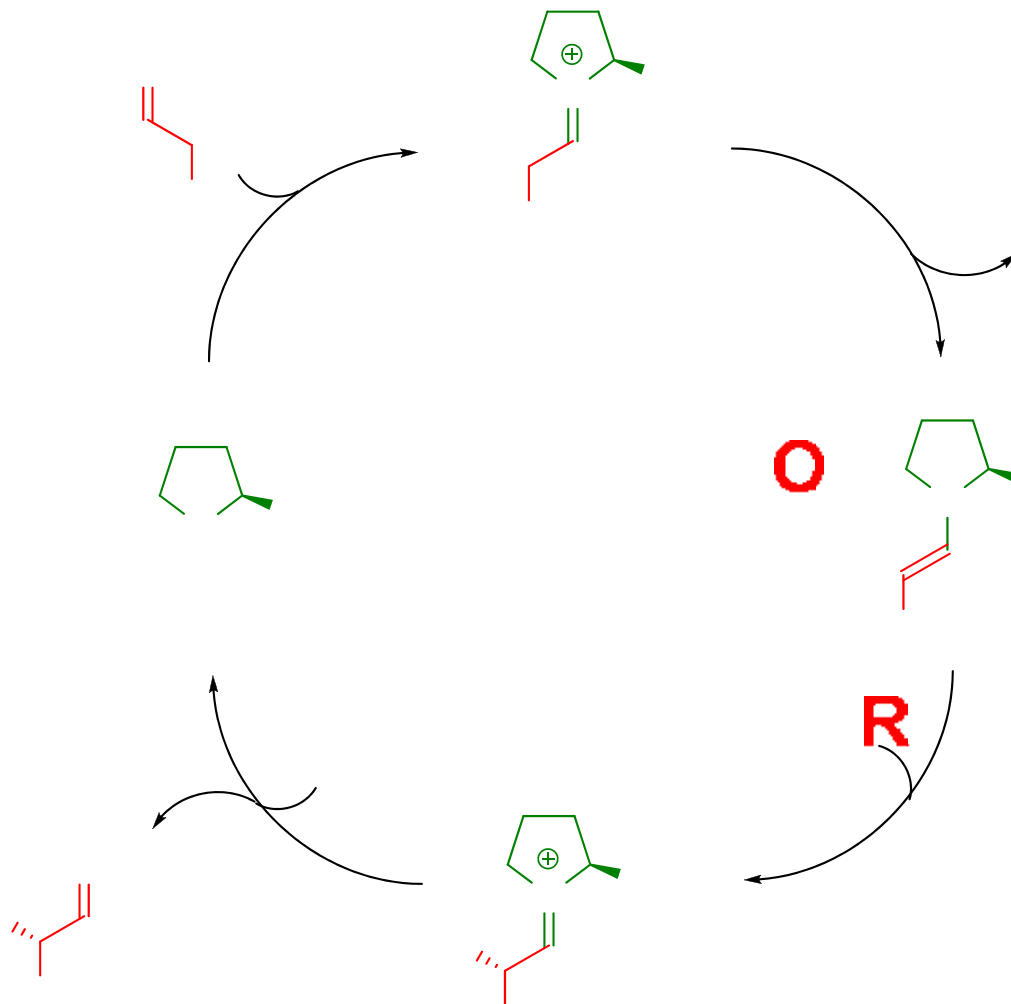
84% yield
78% ee



81% yield
85% ee

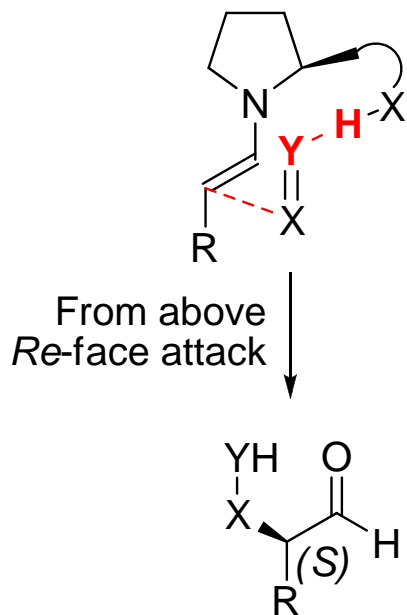
Aldehyde activation

Aldehydes

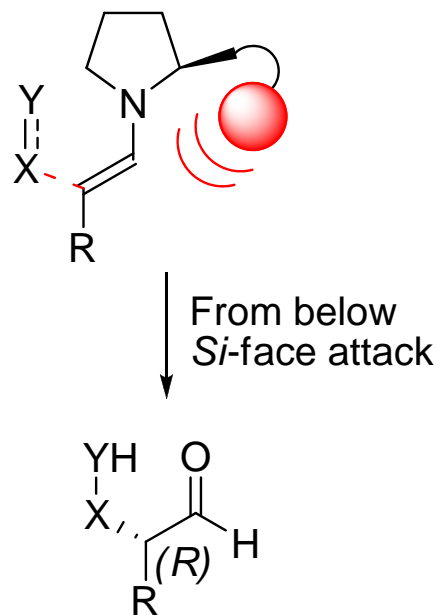


H-Bonding directing vs. steric shielding

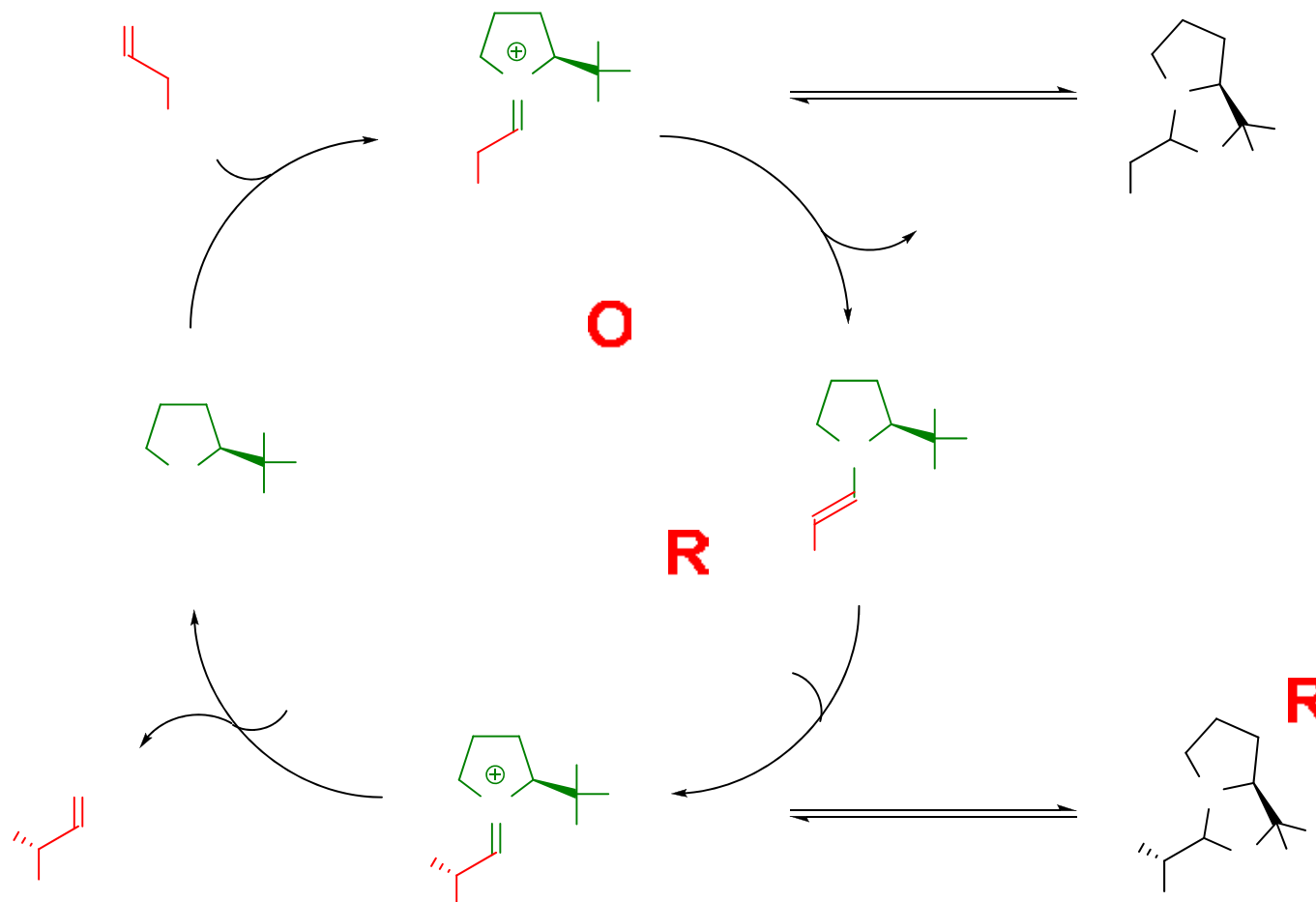
H-bonding directing



Steric shielding



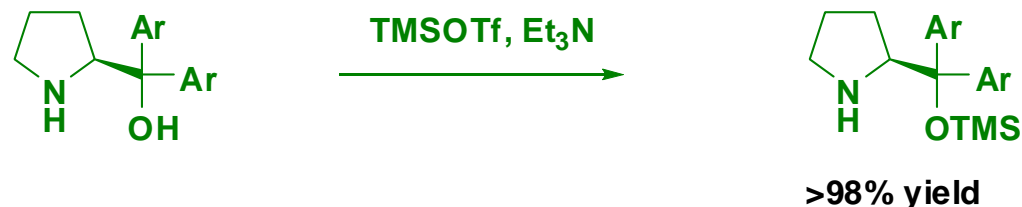
Catalyst design



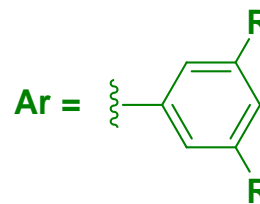
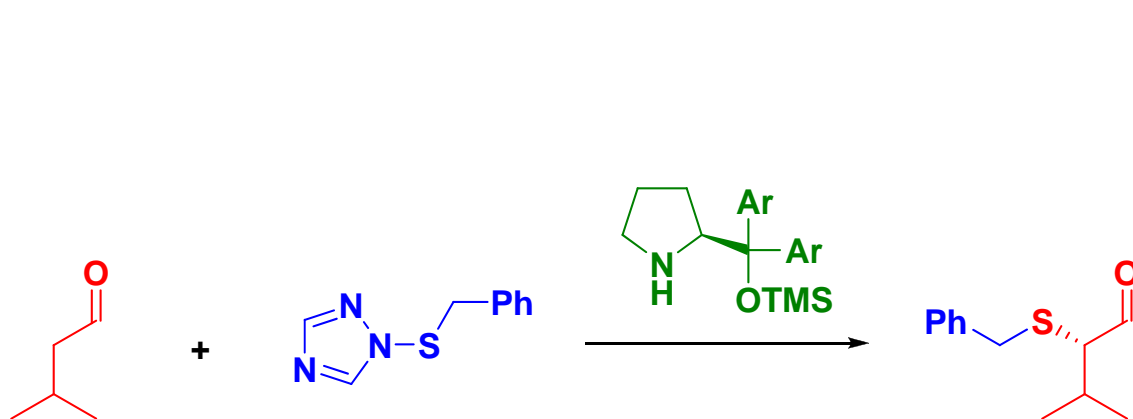
Marigo, Wabnitz, Fielenbach, Jørgensen *Angew. Chem. Int. Ed.* 2005, 44, 792
Franzén, Marigo, Fielenbach, Wabnitz, Kjærsgaard, Jørgensen *J. Am. Chem. Soc.* 2005, 127, 18296
Bertelsen, Jørgensen, *Chem. Soc. Rev.* 2009 Advanced article

Ph

Catalyst synthesis and properties

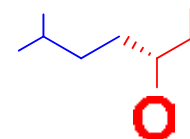
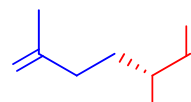
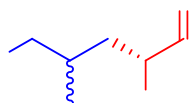
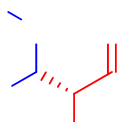
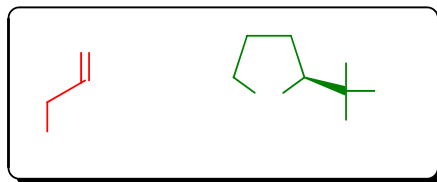


1. The TMS-group can be installed in one single quantitative step.
2. The catalyst is soluble in the most common organic solvents.
3. The reaction can be easily followed by ^1H NMR spectroscopy.

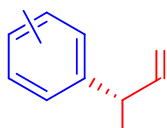


R	Taft E's Value	ee [%]
H	0	77
CH ₃	- 1.24	90
CF ₃	- 2.40	98

α -Functionalization of aldehydes



+



PMP



NH

O



O₂N

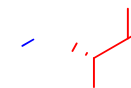


O

EtO₂C



R



R

98% ee

JACS 2005, 127, 18286

85% ee

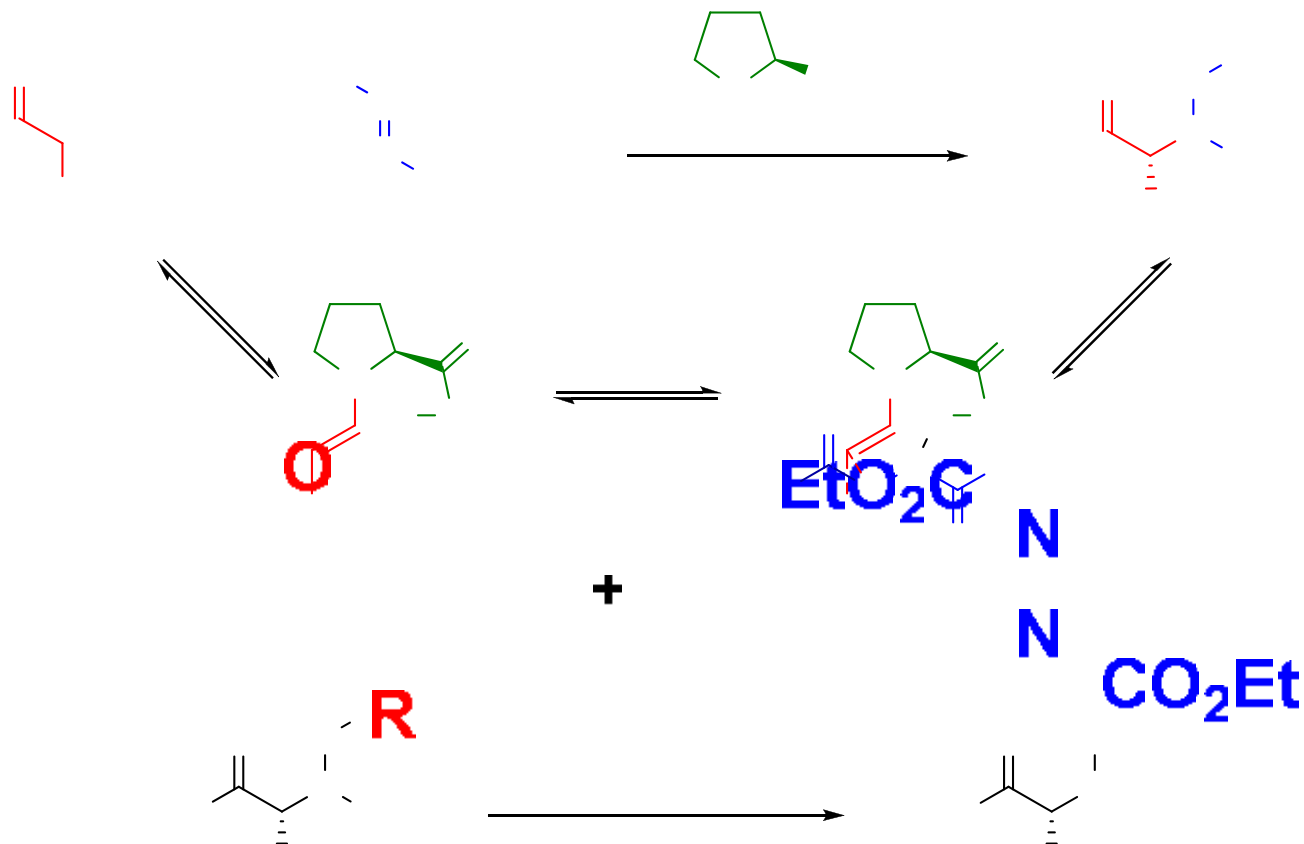
Hayashi: ACIE 2005, 44, 4212

Feature article: Marigo, Jørgensen *Chem. Commun.* 2001, 2001
Bertelsen, Nielsen, Jørgensen *Angew. Chem. Int. Ed.* 2007, 46, 1356

98% ee

Gellman: JACS 2008, 130, 5608

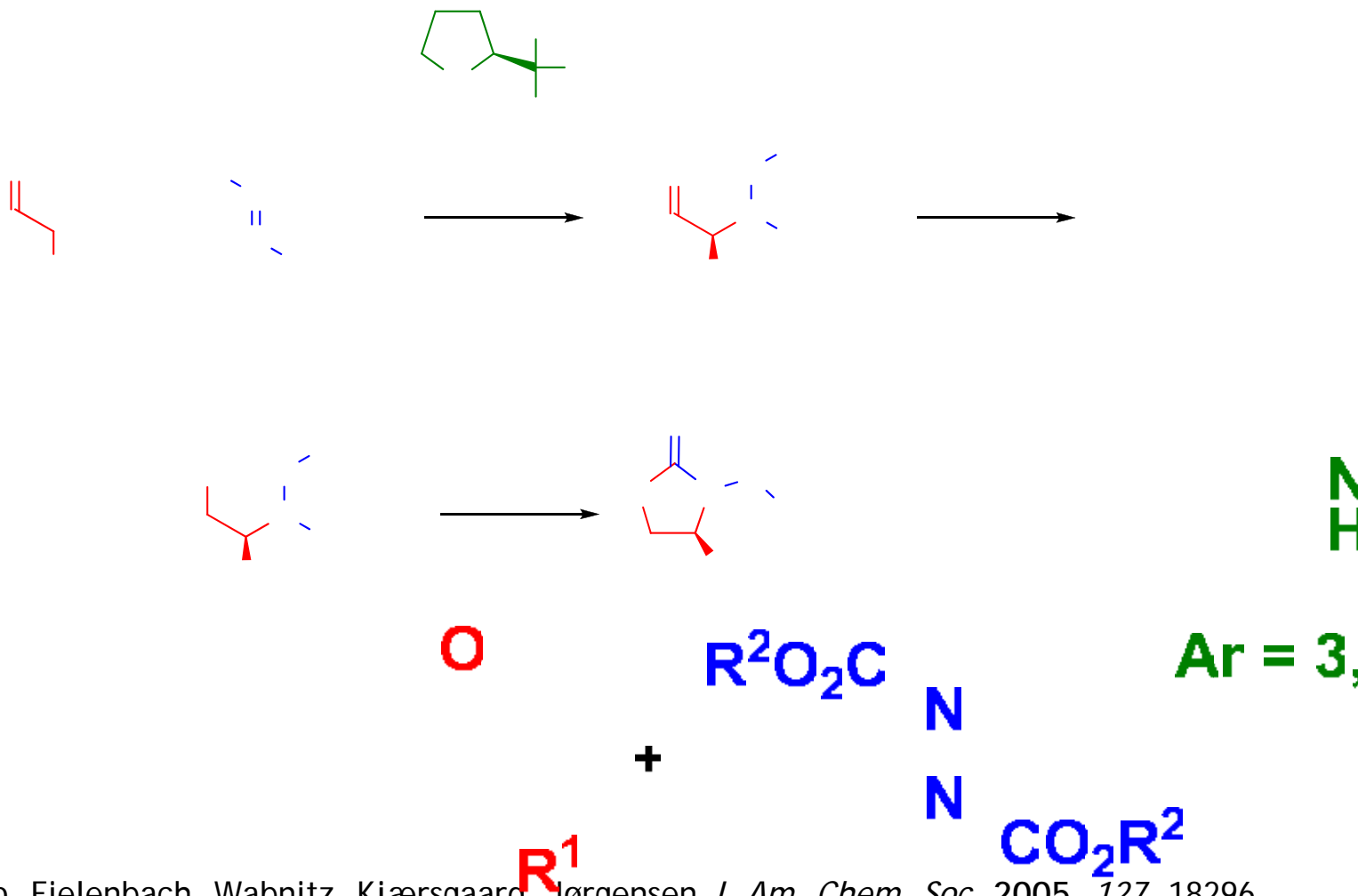
α -Amination reactions



Bøgevig, Kumagurubaran, Juhl, Zhuang, Jørgensen *Angew. Chem. Int. Ed.* 2002, 41, 1790
 List *Am. Chem. Soc.* 2002, 124, 5656
 Ketones: Kumagurubaran, Juhl, Zhuang, Bøgevig, Jørgensen *J. Am. Chem. Soc.* 2002, 124, 6254 14

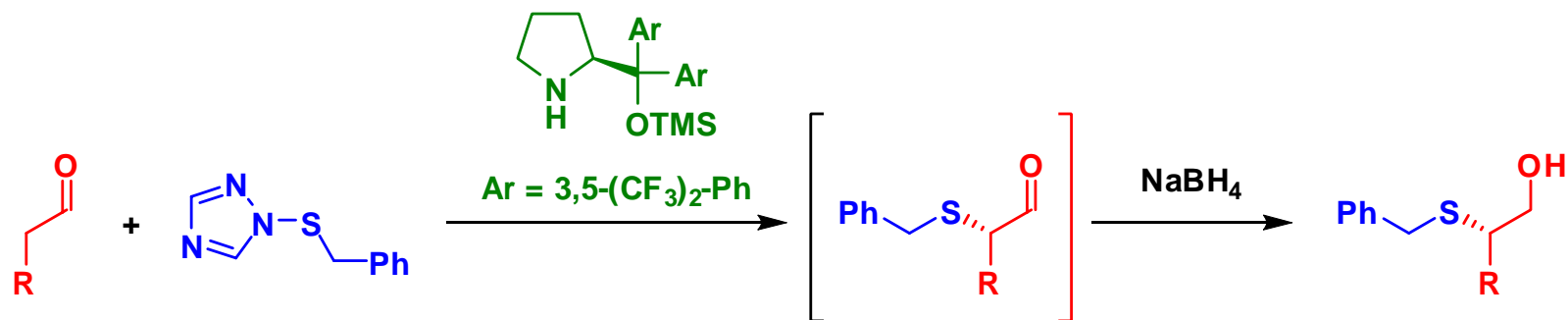
O
 N
 H O

New catalyst and scope



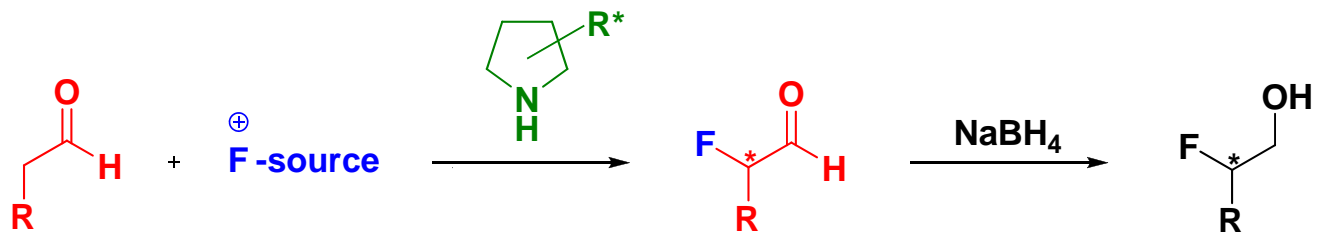
Franzén, Marigo, Fielenbach, Wabnitz, Kjærsgaard, Jørgensen *J. Am. Chem. Soc.* 2005, 127, 18296

α -Sulfenylation reactions



R	Yield [%]	ee [%]
<i>i</i> -Pr	81	98
Me	60	95
Et	85	96
Bn	94	97
Allyl	64	96
<i>t</i> -Bu	83	95

α -Fluorination



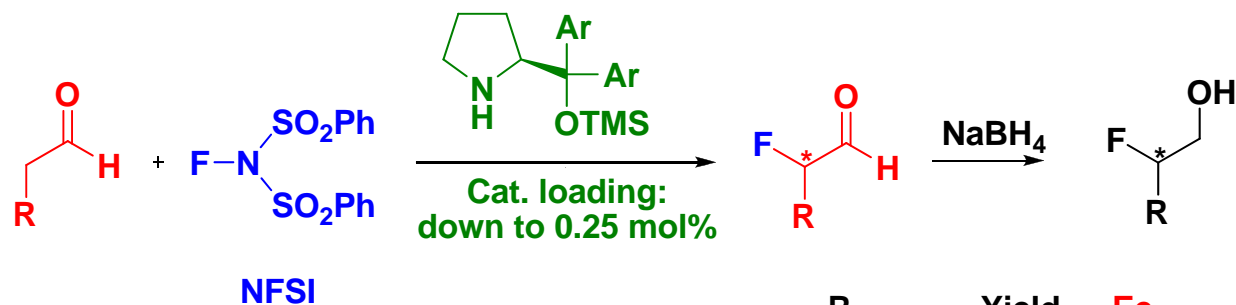
Direct use in pharmaceuticals,
agro and material science

Problem-challenge:

α -Fluoro aldehydes are unstable and very volatile -
racemization - the high electronegativity of F

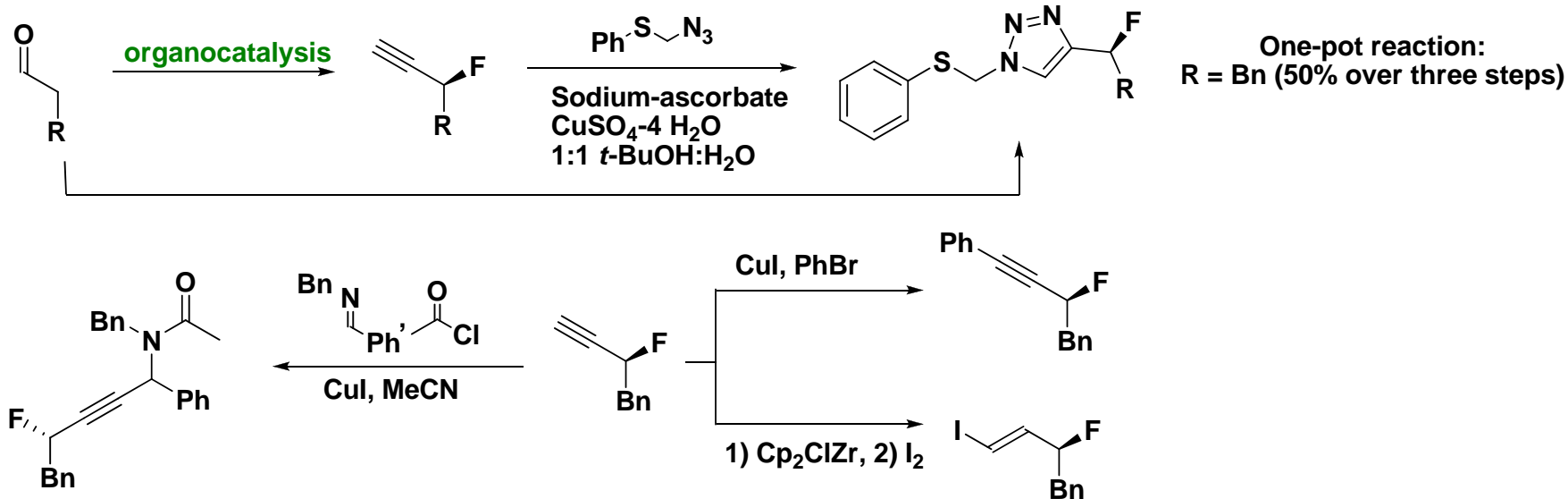
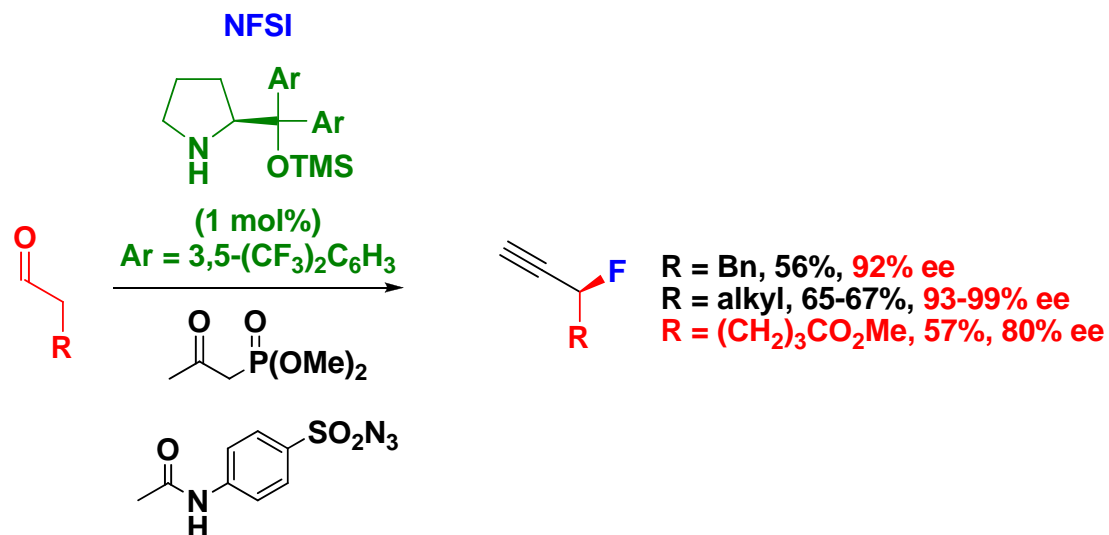
Enders, Hüttl *Synlett*, 2005, 991
Marigo, Fielenbach, Branton, Kjærsgaard, Jørgensen *Angew. Chem. Int. Ed.* 2005, 44, 3703
Steiner, Mase, Barbas *Angew. Chem. Int. Ed.* 2005, 44, 3706
Beeson, MacMillan *J. Am. Chem. Soc.* 2005, 127, 8826

Scope of the reaction

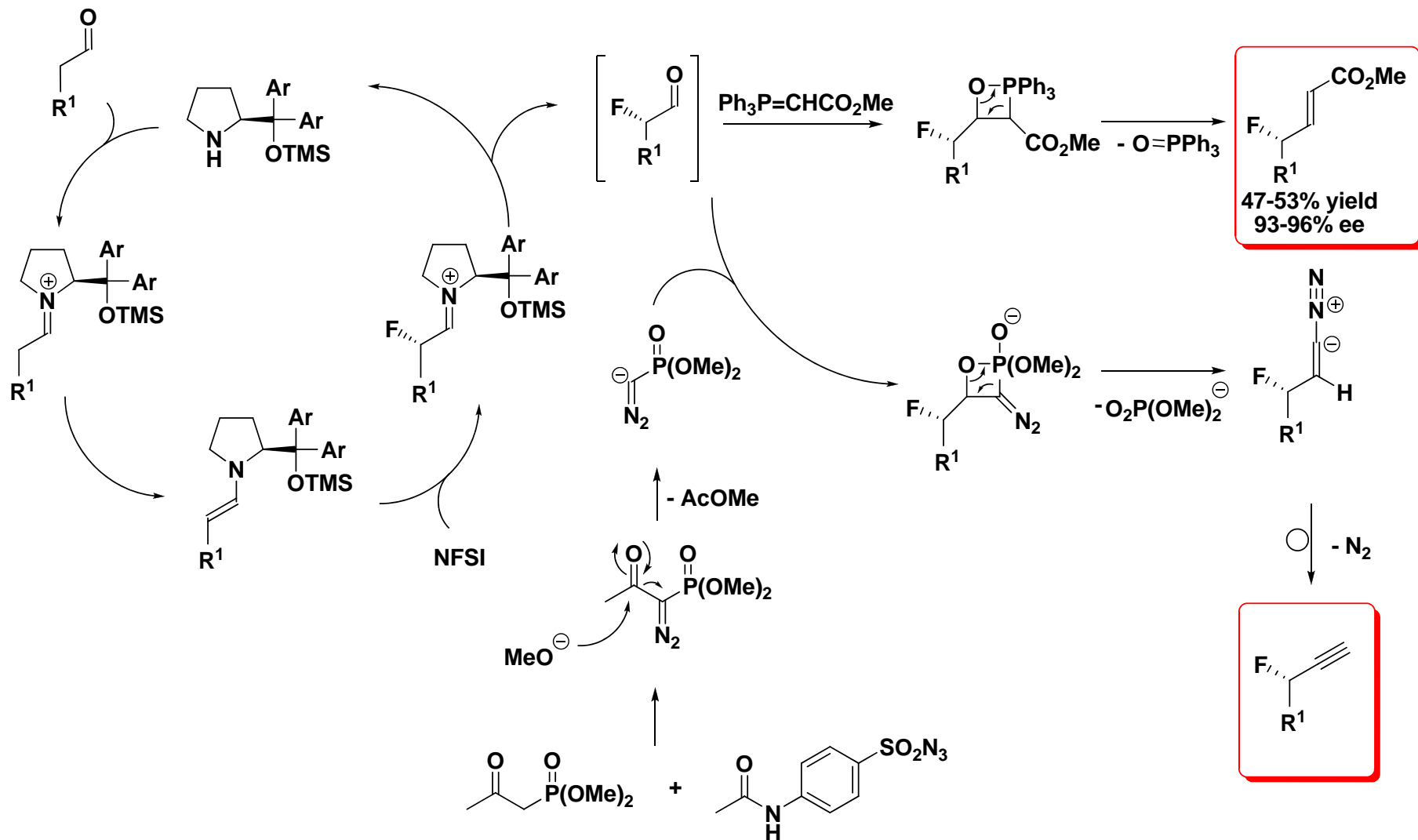


<u>R</u>	<u>Yield</u>	<u>Ee</u>
Pr	>95	96
Bu	>90	91
Hex	55	96
Bn(CH ₂) ₃	64	91
Bn	74	93
Cy	69	96
<i>t</i> -Bu	>90	97
1-Ad	75	96

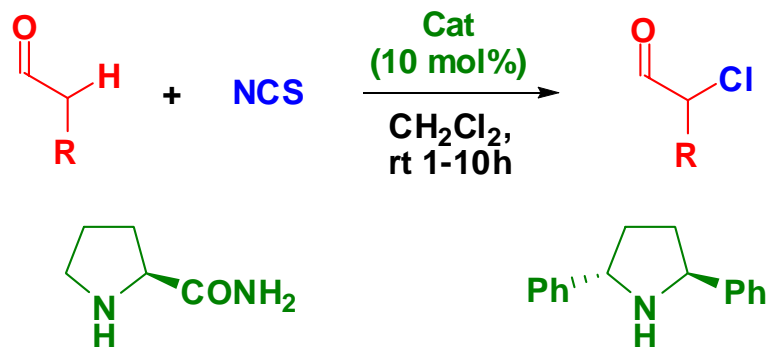
Propagylc fluorides



Mechanism

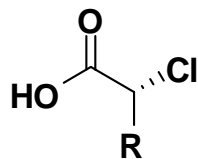
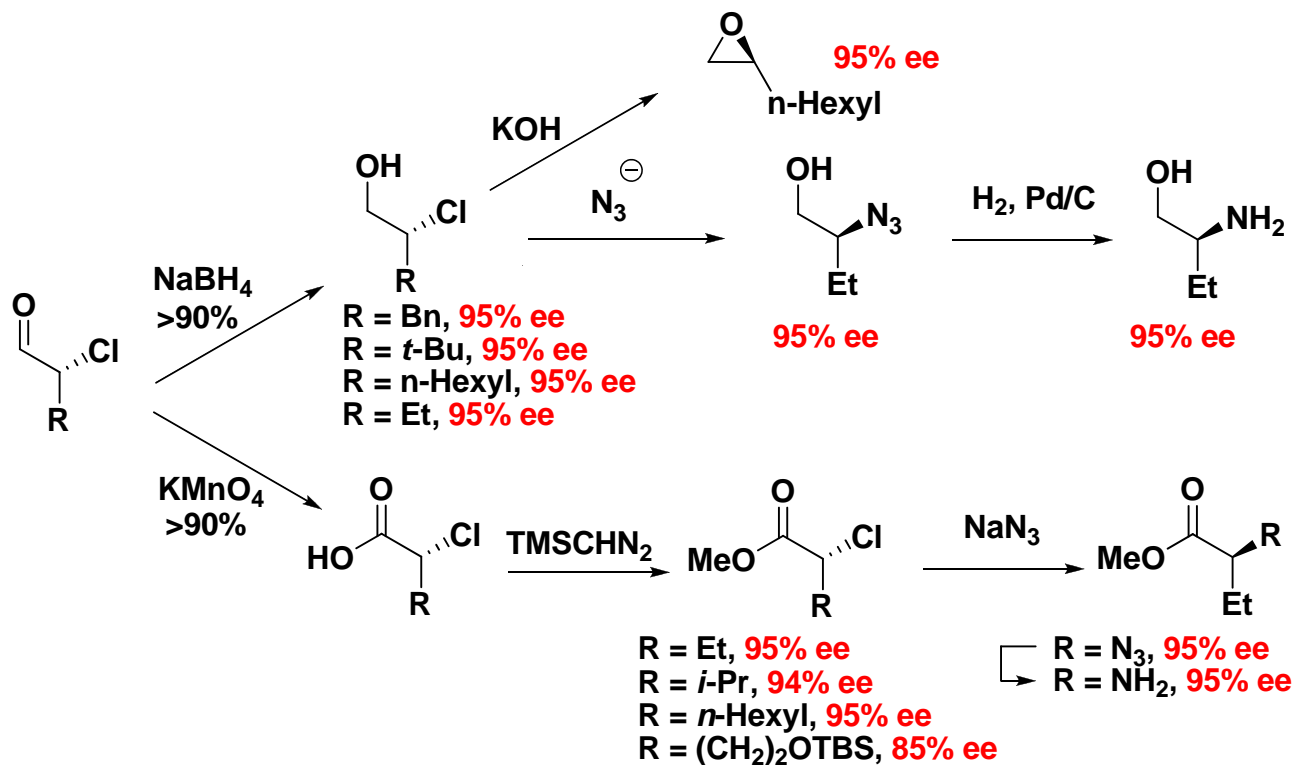


Variation in aldehydes



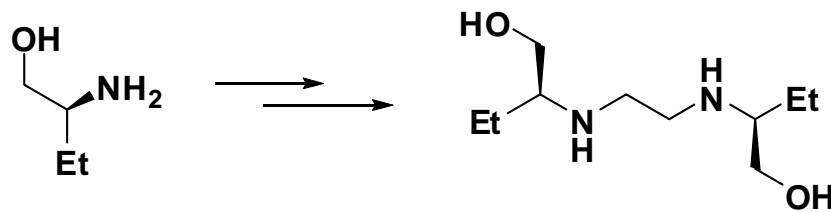
R	Yield	Ee	Yield	Ee
Me	99	75	-	-
Et	99	80	90	95
<i>i</i> -Pr	95	87	90	94
<i>t</i> -Bu	93	95	-	-
<i>n</i> -Hex	95	70	99	95
Allyl	90	74	90	95
CH ₂ Ph	75	78	82	95
(CH ₂) ₂ OTBS	75	85	95	81

Synthetic manipulations



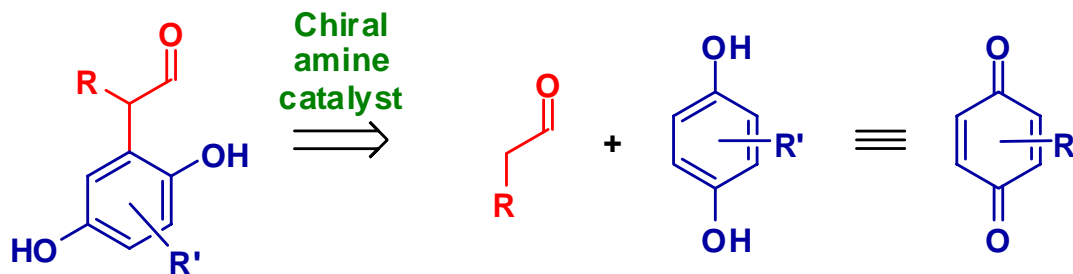
Prepared in multi-tons scale/year

One of the most important amino alcohols
ACIE 2004, 43, 788

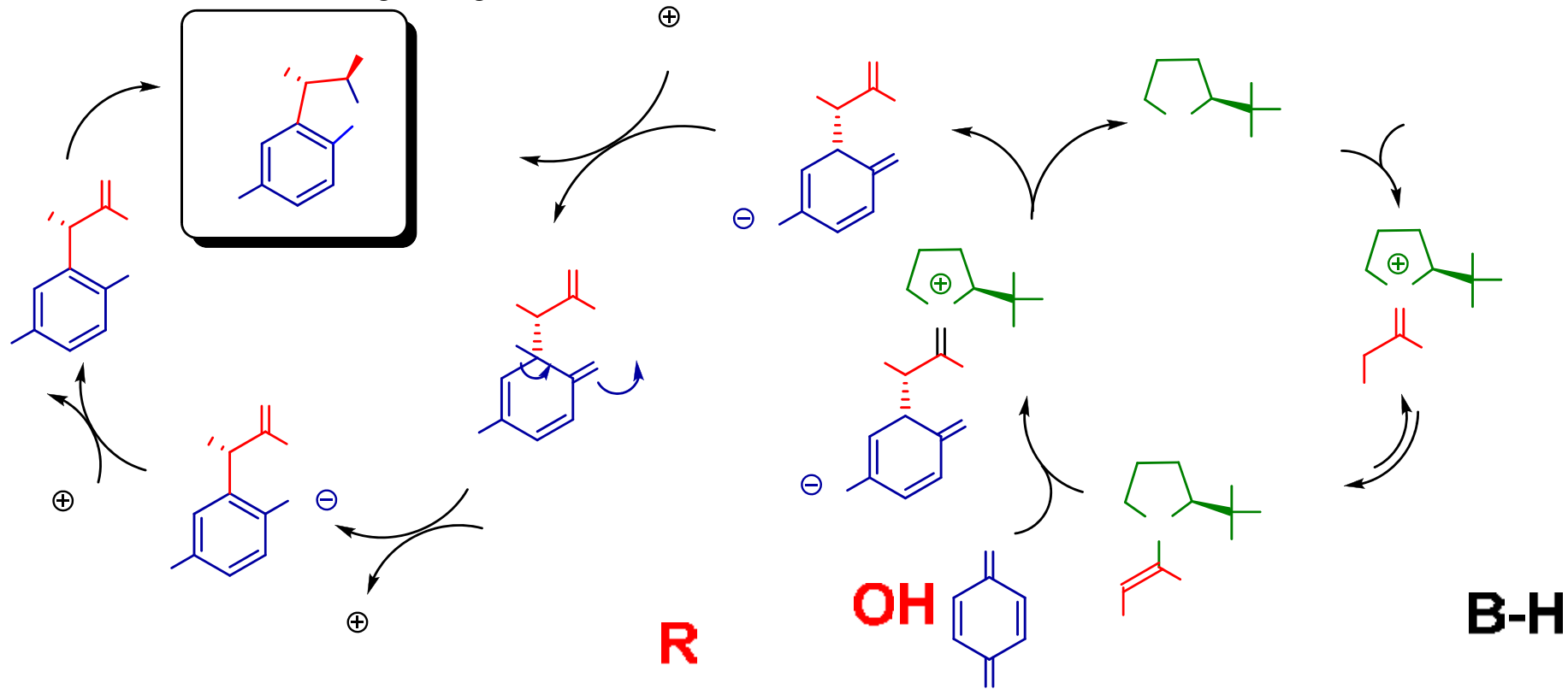


Tuberculostatic
etambutol

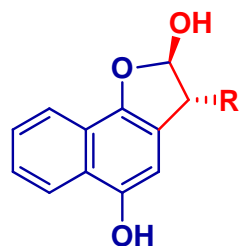
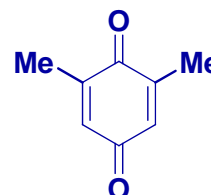
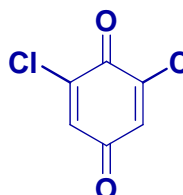
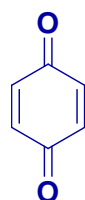
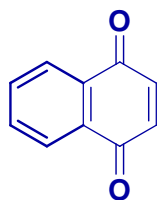
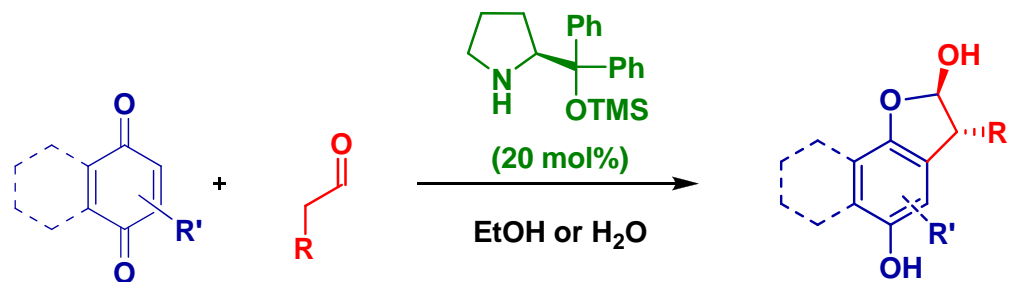
α -Arylation



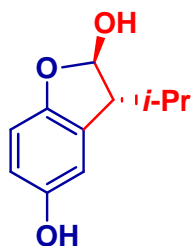
Mechanism - two catalytic cycles



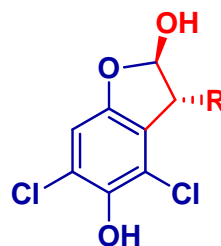
α -Arylation of aldehydes



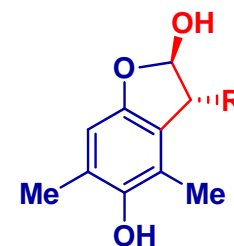
R = alkyl
86% - quant yield
96 - >99% ee



52% yield
99% ee

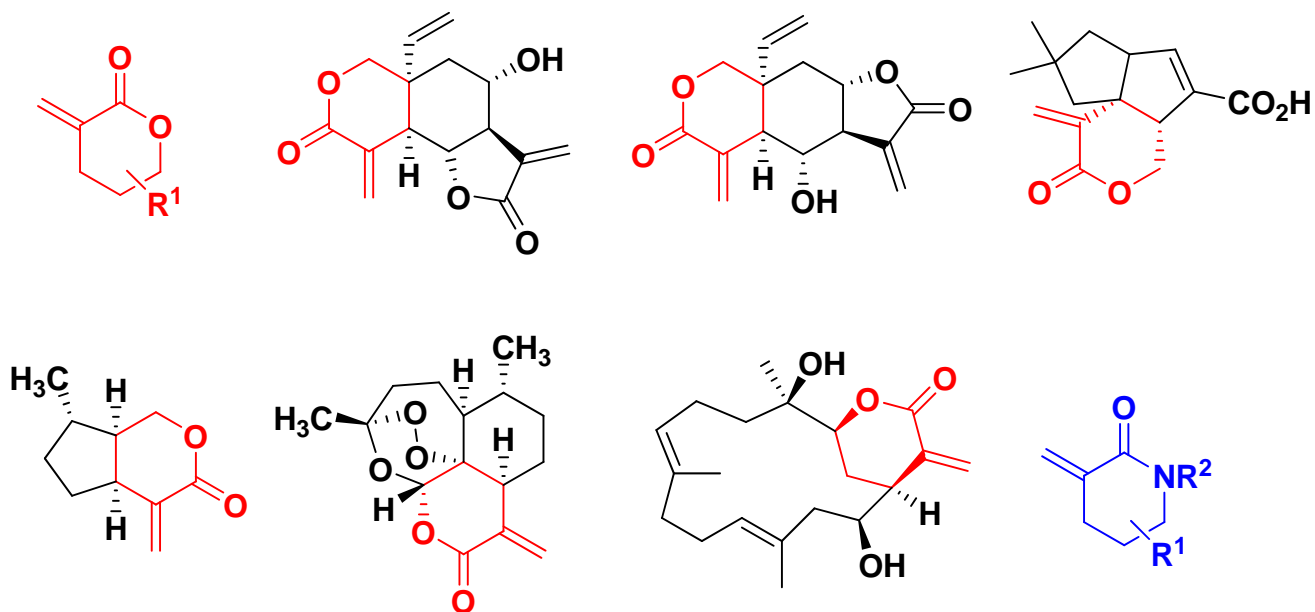


R = alkyl
65-95% yield
96 - >99% ee

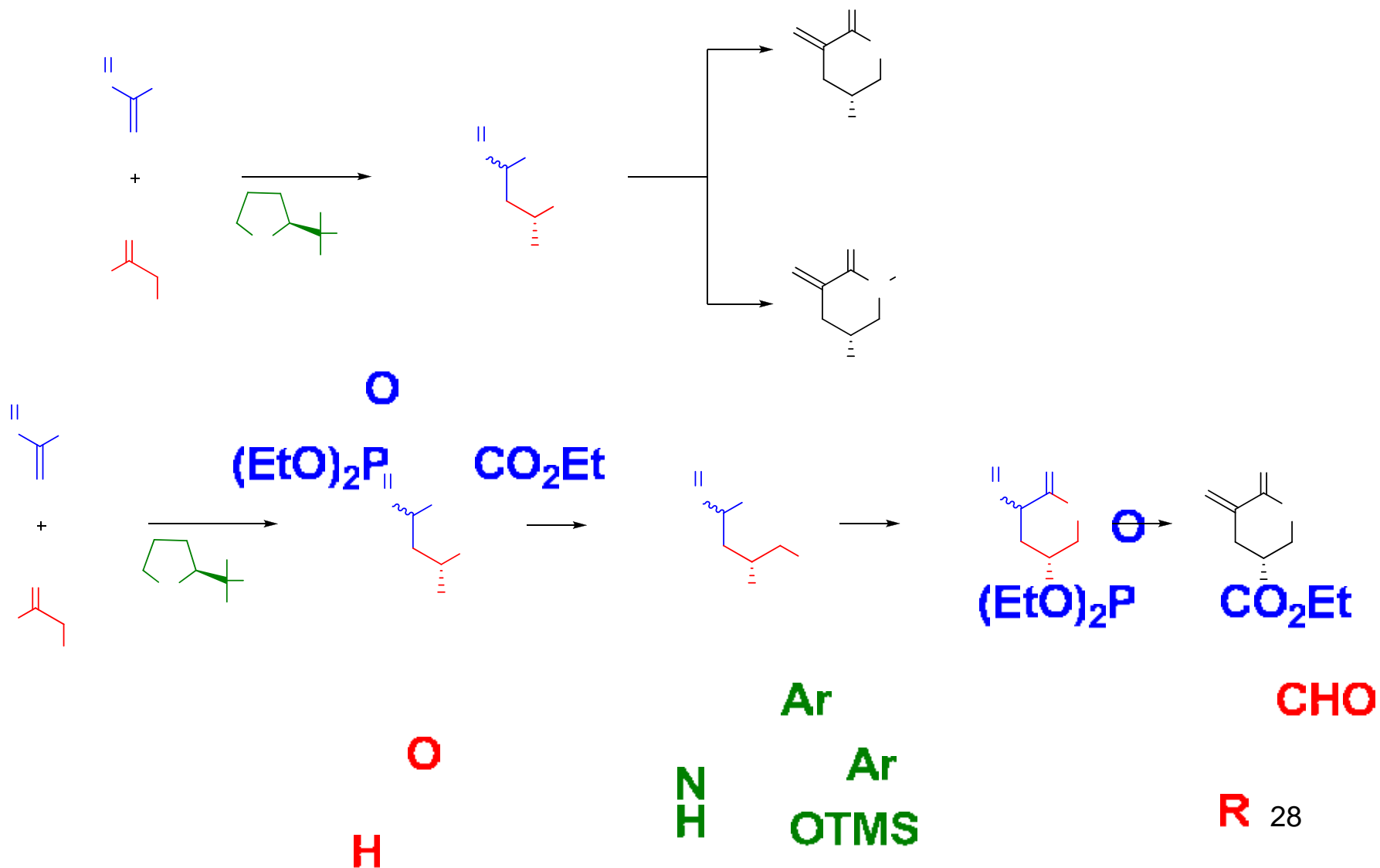


R = alkyl
72-85% yield
>98% ee

Application - α -methylene- δ -lactones and δ -lactams



α -Methylene- δ -lactones and δ -lactams

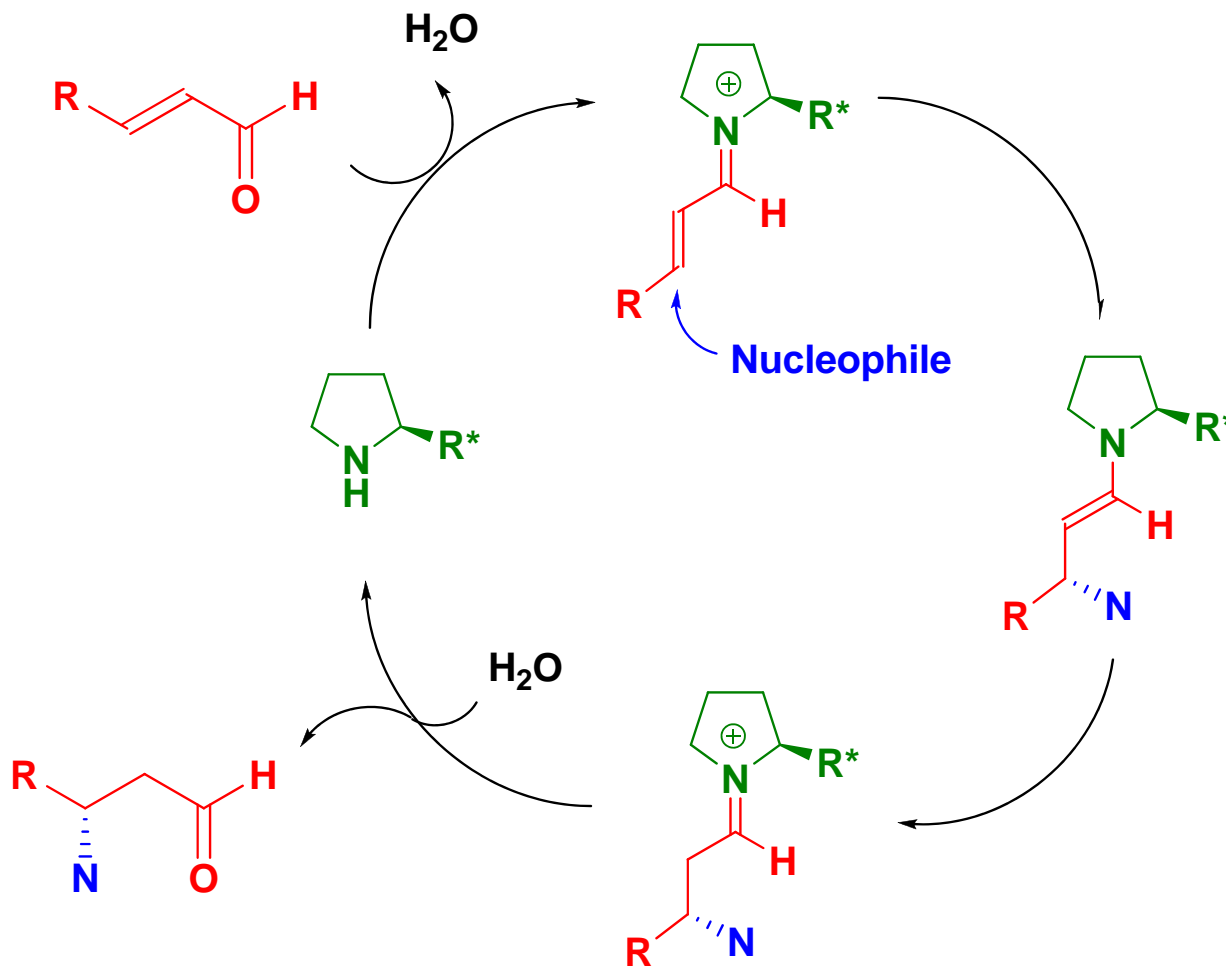


From α - to β -functionalization

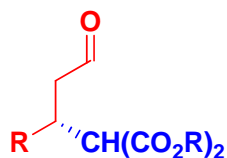
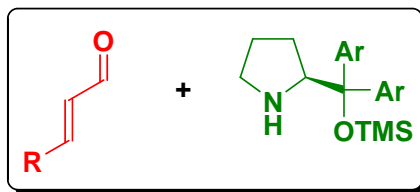


Aldehyde activation

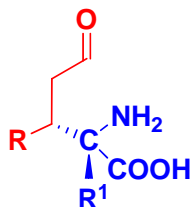
α,β -Unsaturated aldehydes



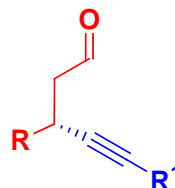
β -Functionalization of α,β -unsaturated aldehydes



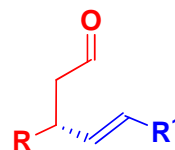
95% ee
ACIE 2006, 45, 4305



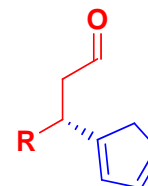
96% ee
JACS 2008, 130, 12031



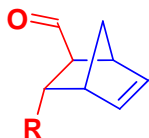
97% ee
JACS 2009,
131, in press



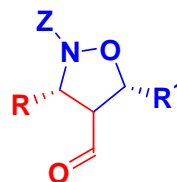
>20:1 (E):(Z); 97% ee
JACS 2009, 131, in press



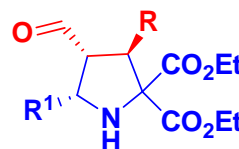
95% ee
Hayashi: ACIE 2006, 45, 6853



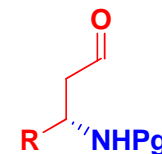
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Hayashi: *OrgLett* 2007, 9, 2859



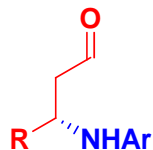
96% ee
Chow: *TetLett* 2007, 48, 277
Cordova: *TetLett* 2007, 48, 5701



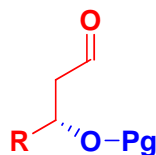
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Cordova: *TetLett* 2007, 48, 6252



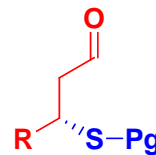
98% ee
Cordova: *ChemComm* 2007, 849



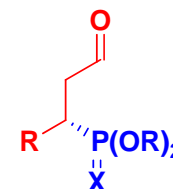
94% ee
ACIE 2007, 46, 1983



97% ee
JACS 2007, 129, 1536

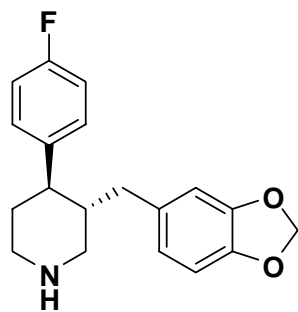


97% ee
JACS 2005, 127, 15710

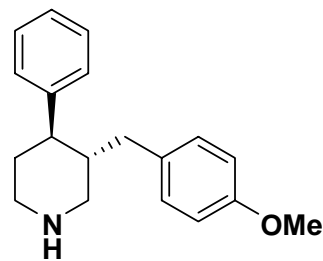


97% ee
Melchiorre: ACIE 2007, 46, 4504
Cordova: ACIE 2007, 46, 4507
JOC 2007, 72, 8893

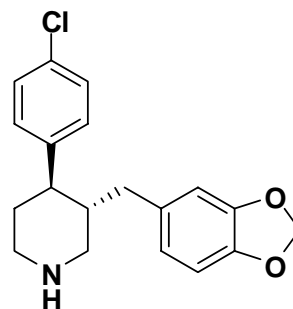
Application



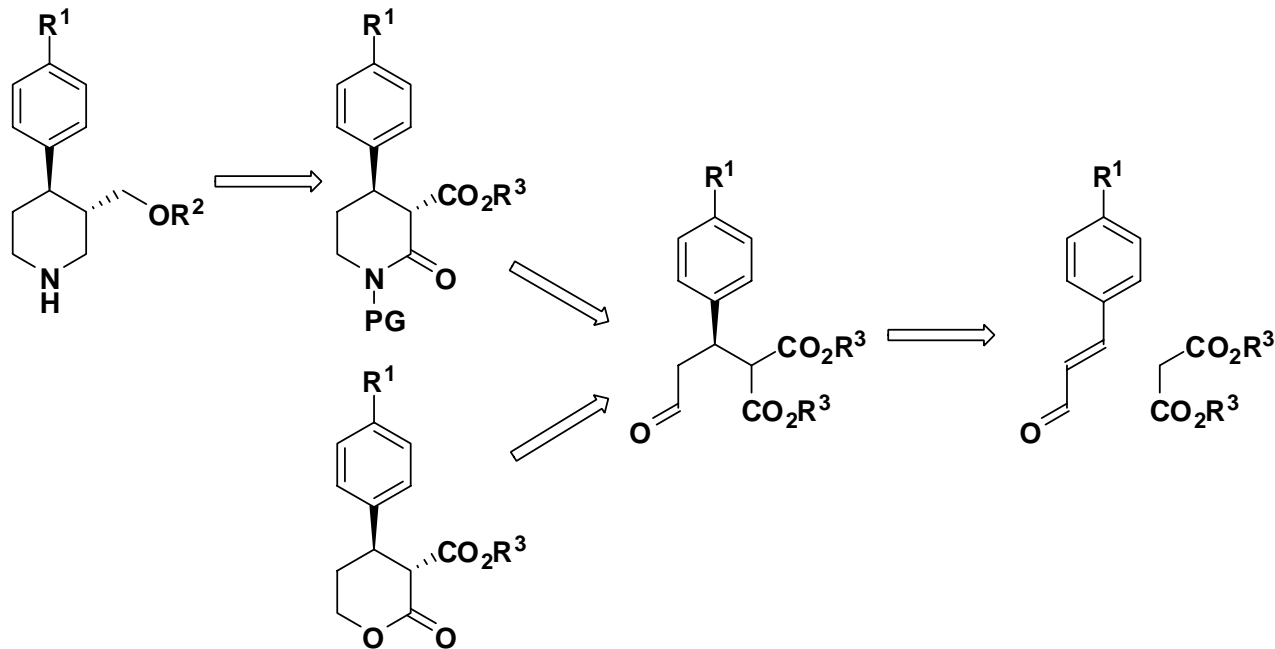
Paroxetine



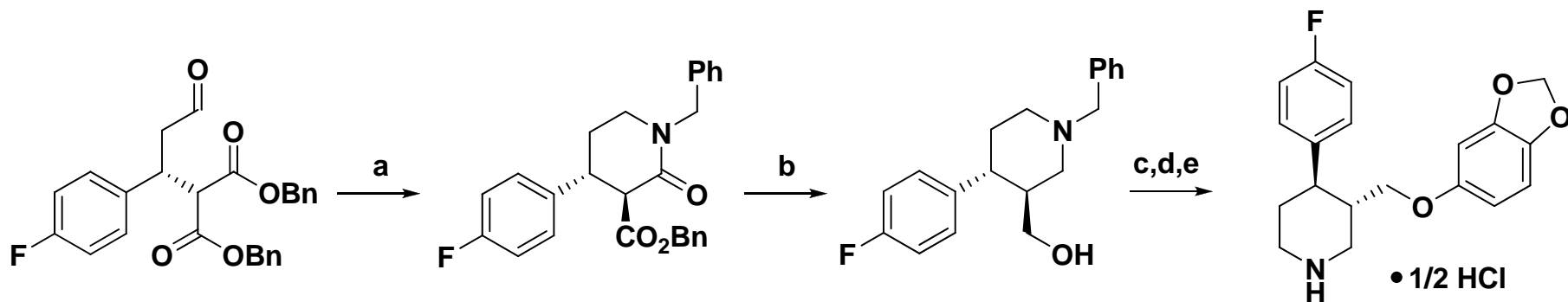
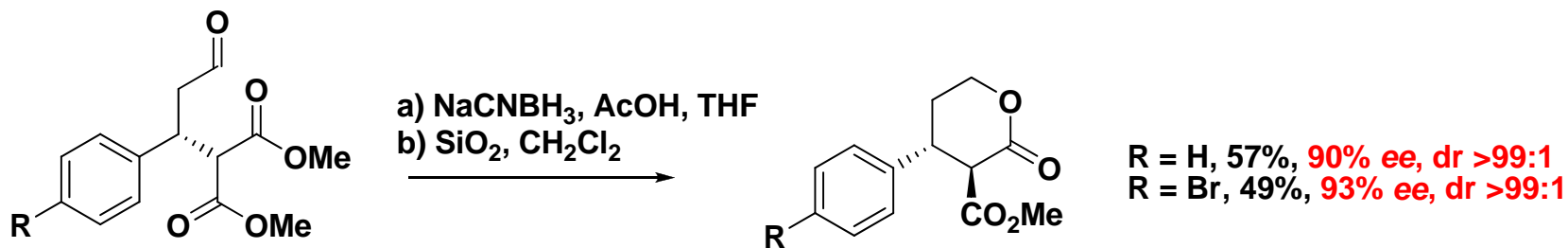
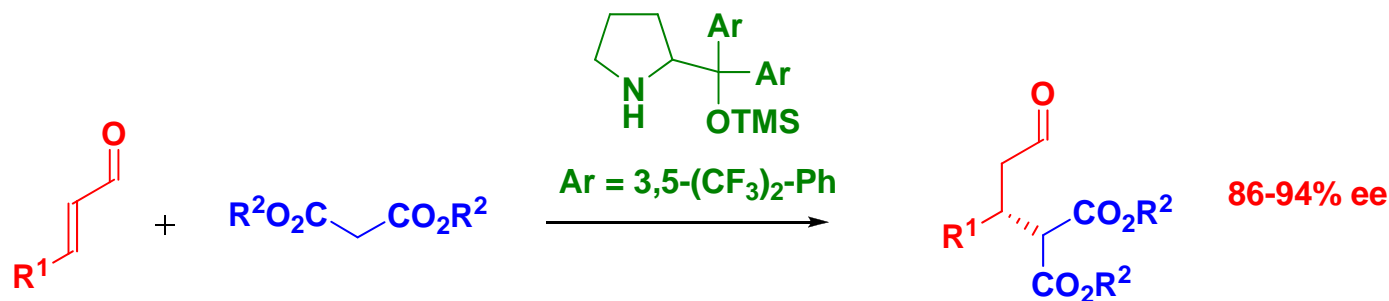
Fermoexetine



Rosche-1
Peptidomimetic inhibitor

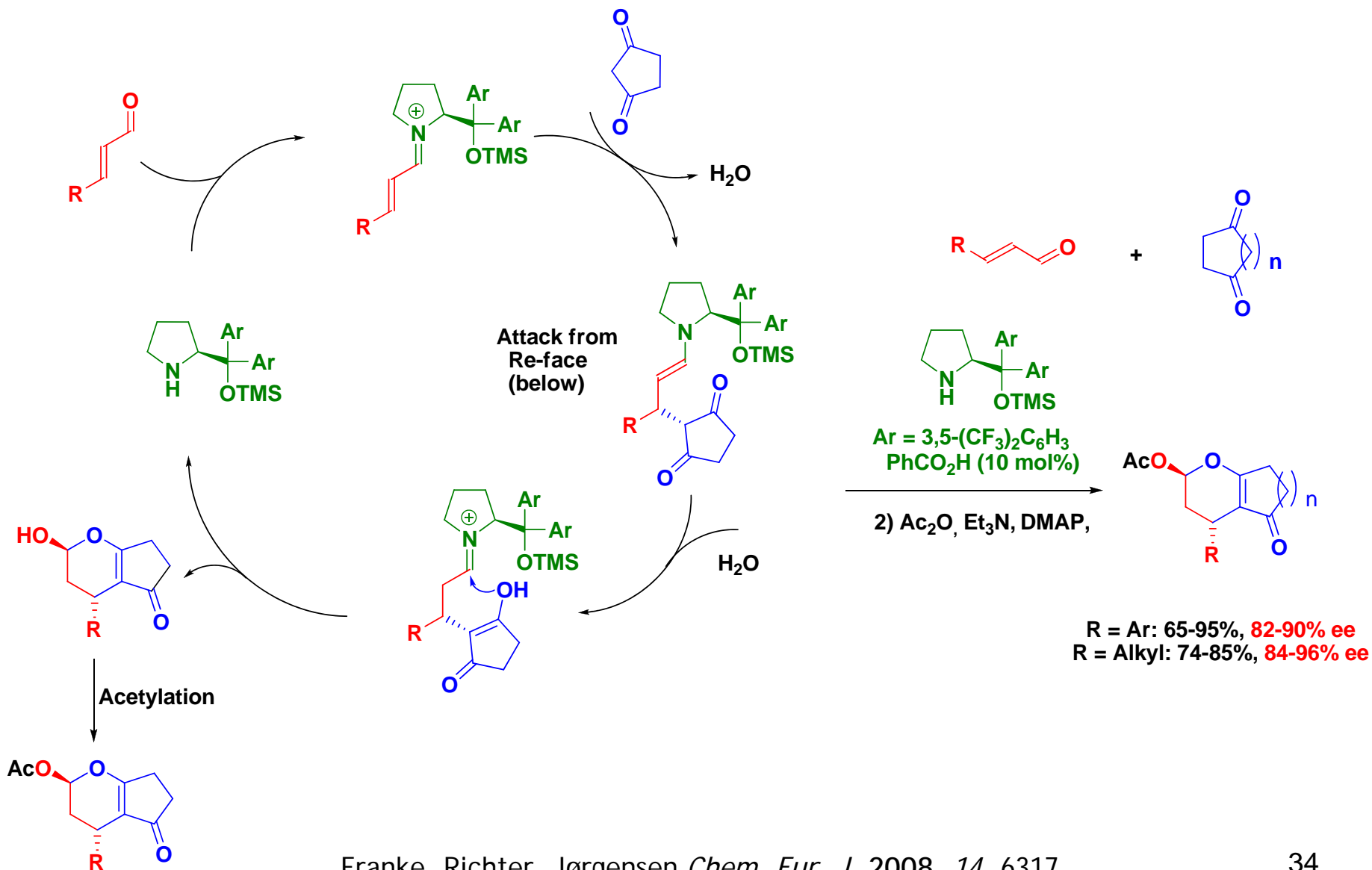


Application

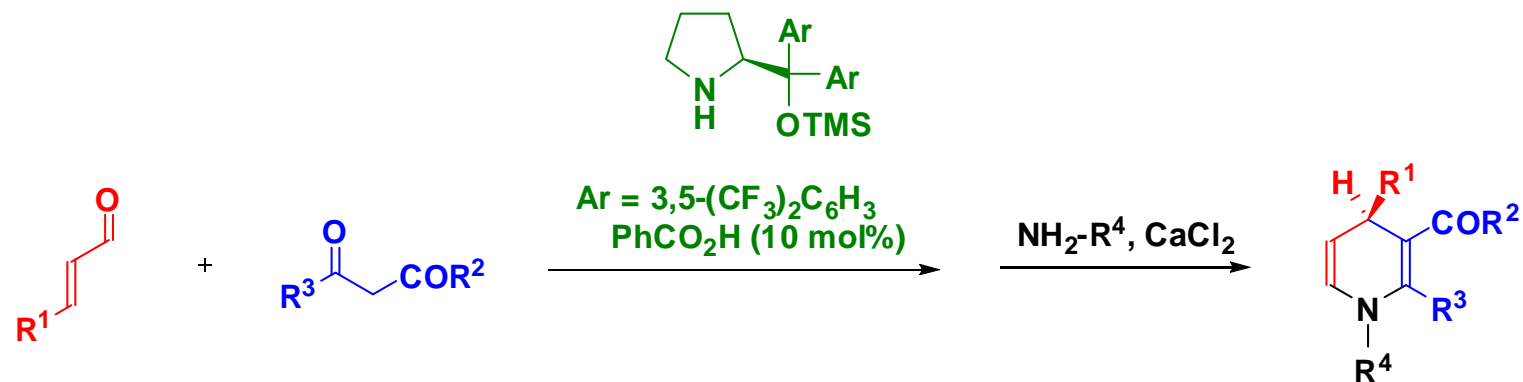


Synthesis of (-)-paroxetine: a) PhCH_2NH_2 , NaBH(OAc)_3 , dioxane, 70%;
 b) LiAlH_4 , THF; c) MsCl , NEt_3 , toluene; d) sesamol, NaH, DMF, 60 °C;
 e) i) H_2 , 5% Pd/C; ii) HCl.

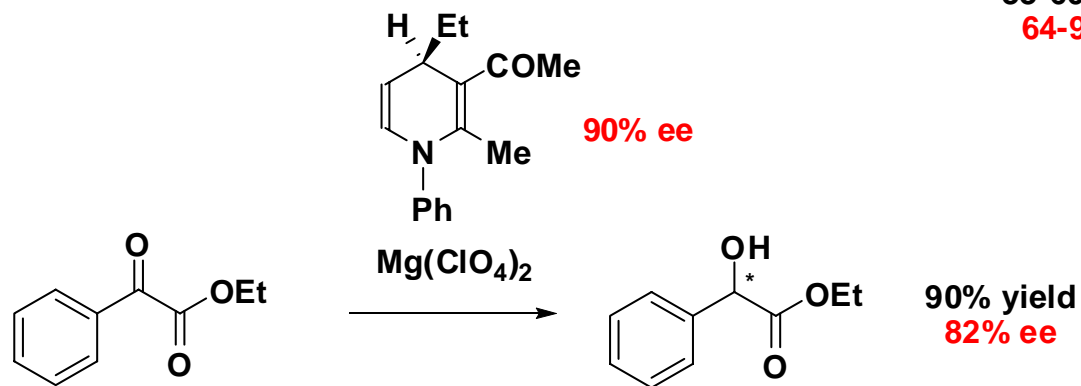
Optically active 3,4-dihydropyran derivatives



Optically active 1,4-dihydropyridines



33-60% yield
64-92% ee

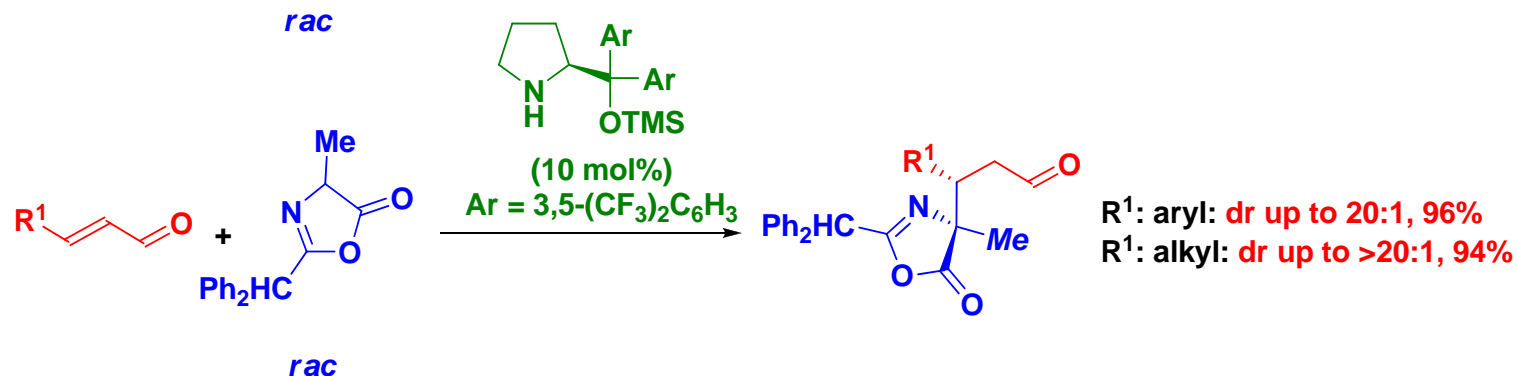
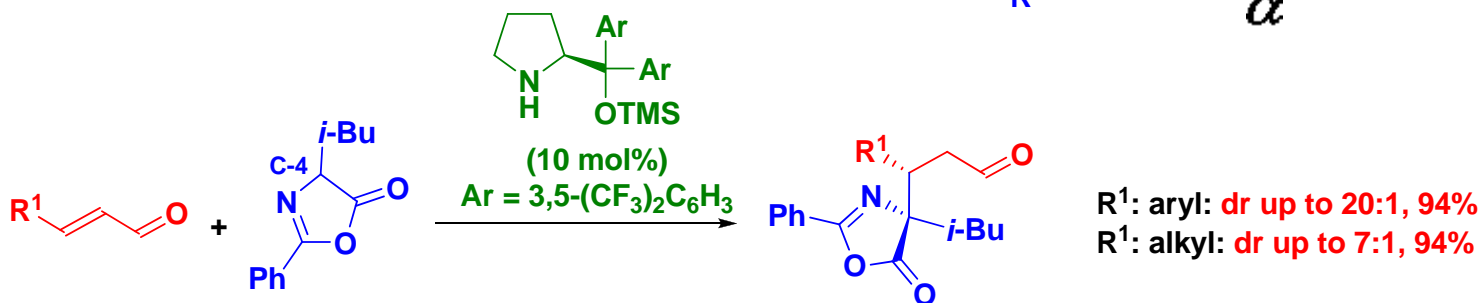
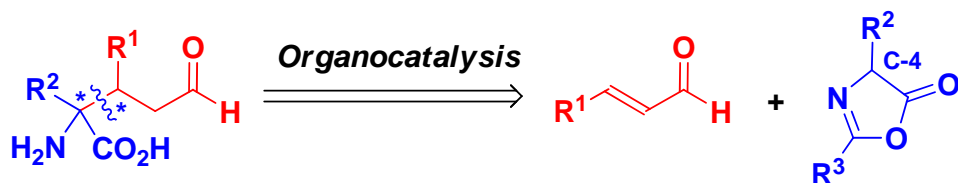
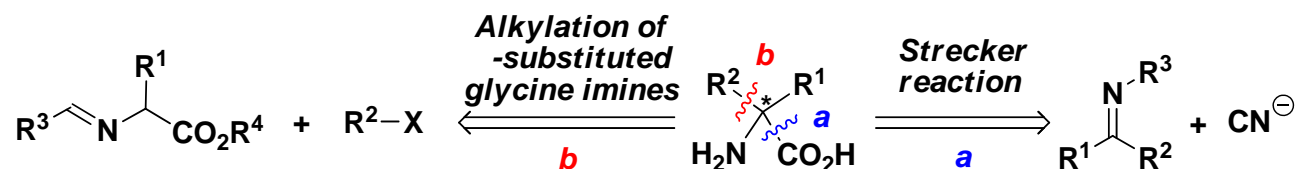


90% ee

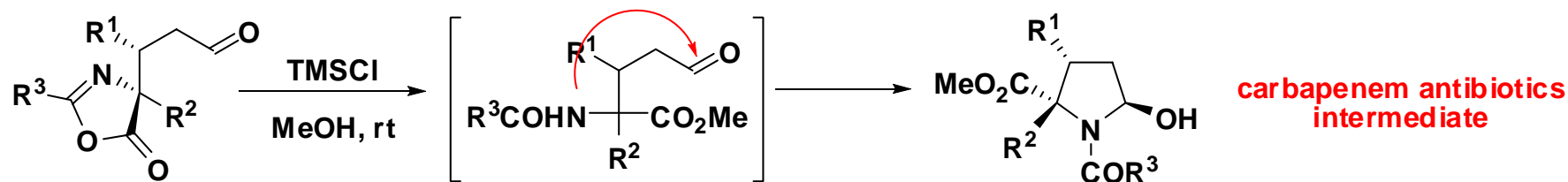
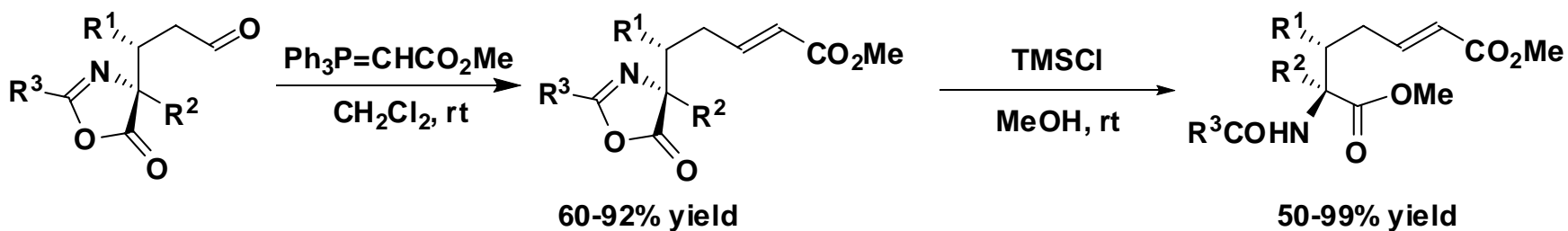
90% yield
82% ee

Franke, Johansen, Bertelsen, Jørgensen *Chem. Asian J.* 2008, 3, 196

Application to amino acid synthesis



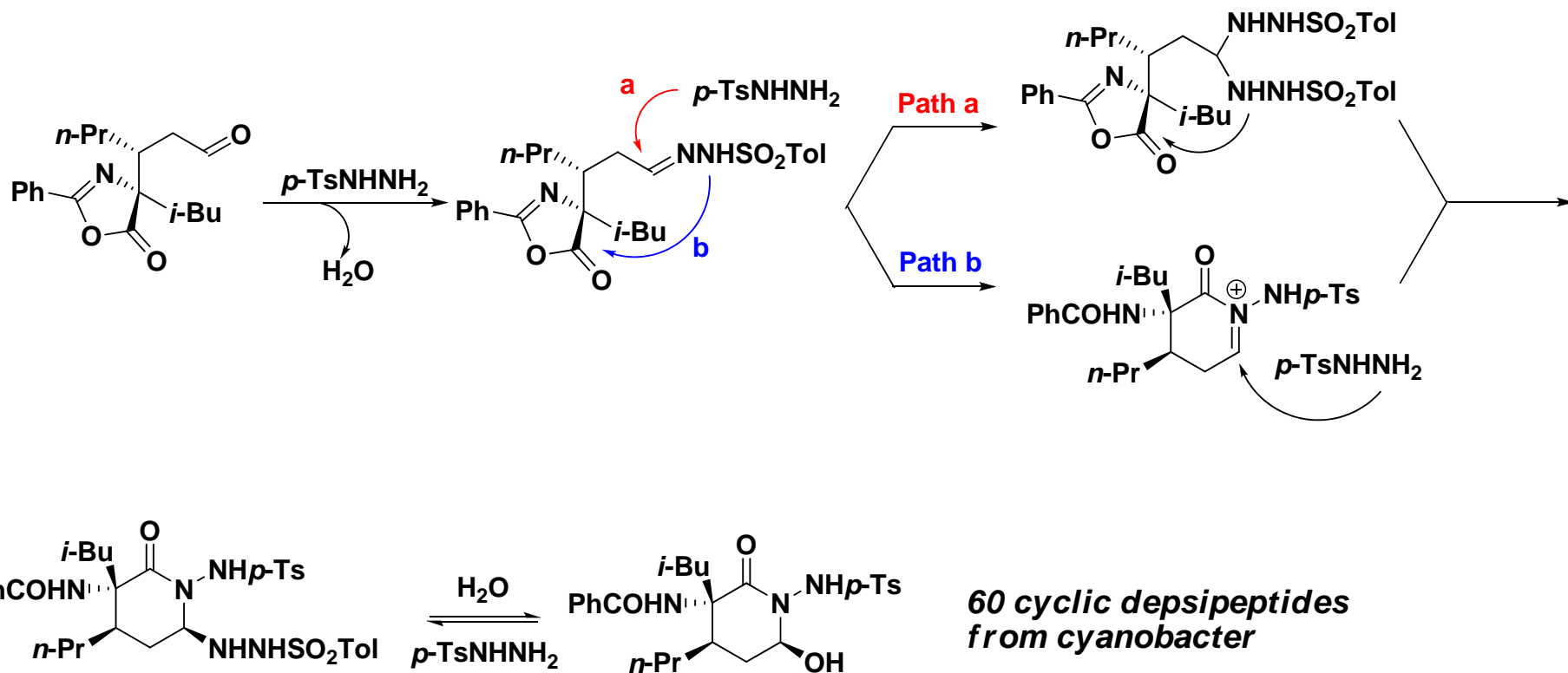
Application to amino acid synthesis



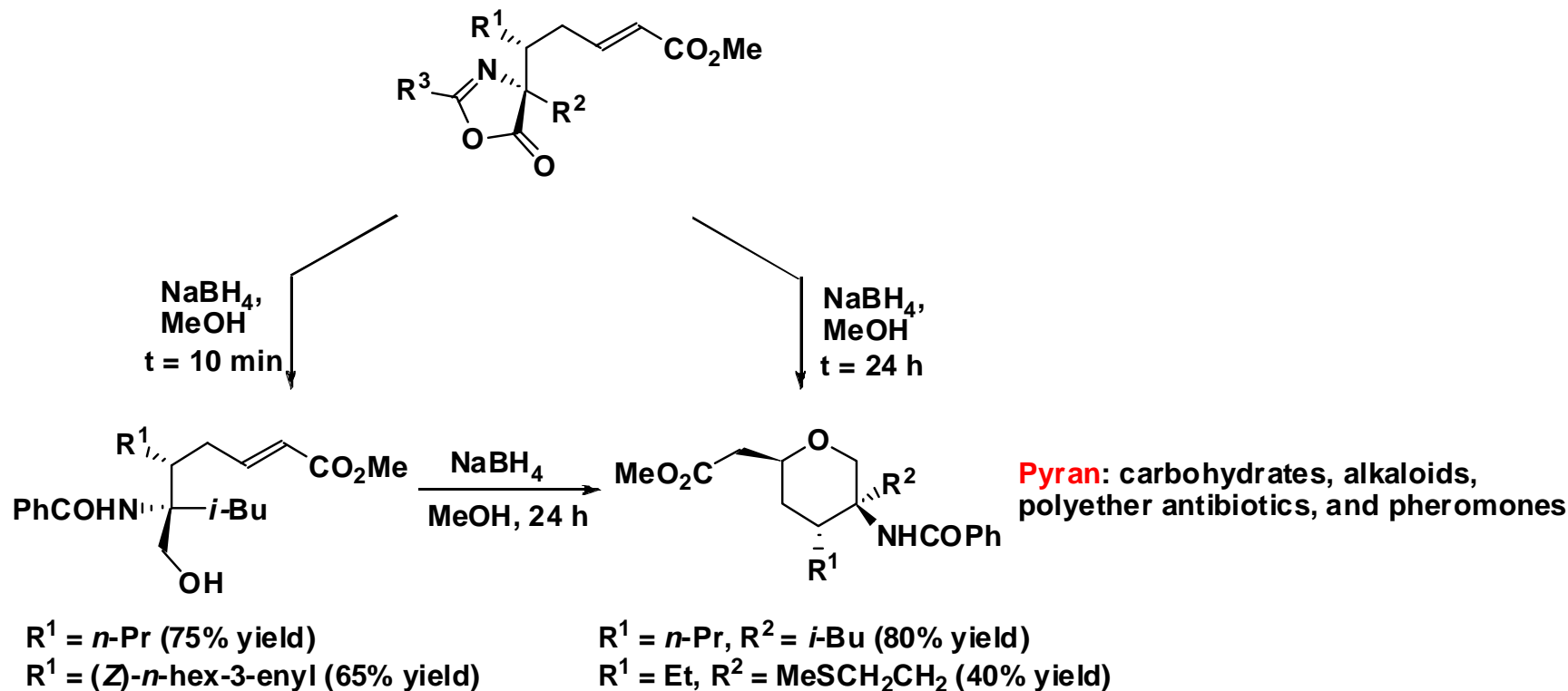
$\text{R}^1 = n\text{-Pr}, \text{R}^2 = i\text{-Bu}, \text{R}^3 = \text{Ph}$ (71% yield)

$\text{R}^1 = \text{Et}, \text{R}^2 = i\text{-Bu}, \text{R}^3 = o\text{-ClC}_6\text{H}_4$ (90% yield)

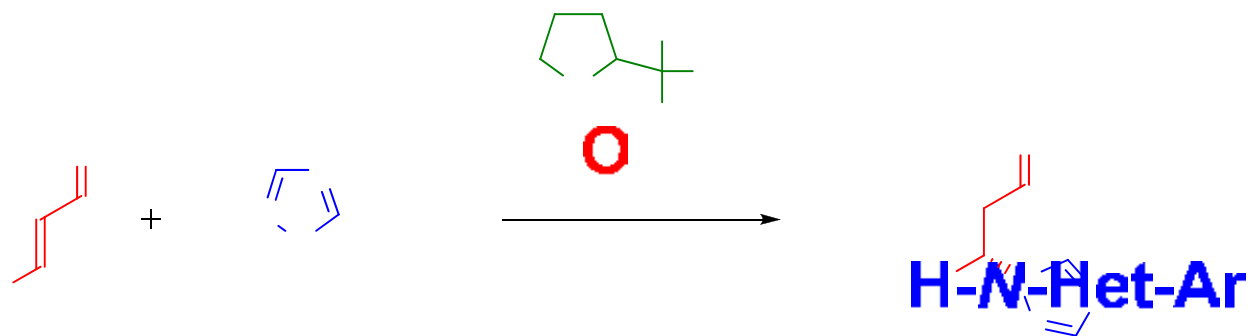
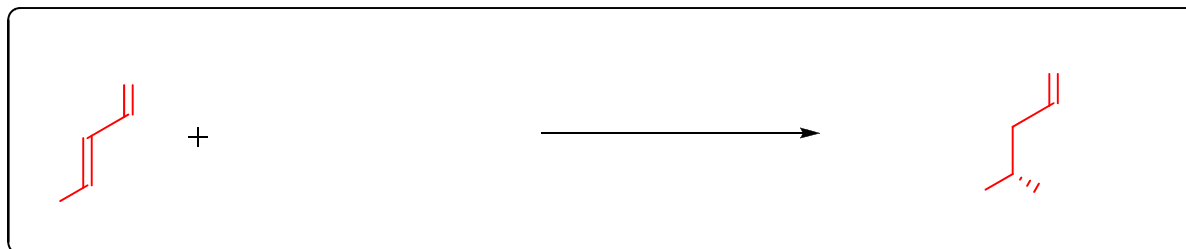
Application to amino acid synthesis



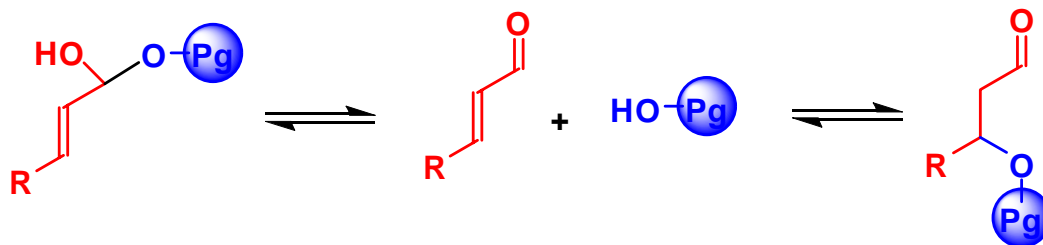
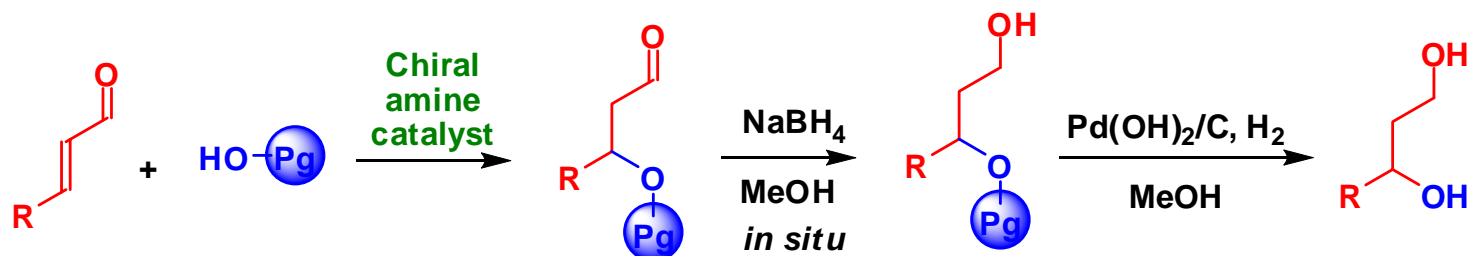
Application to amino acid synthesis



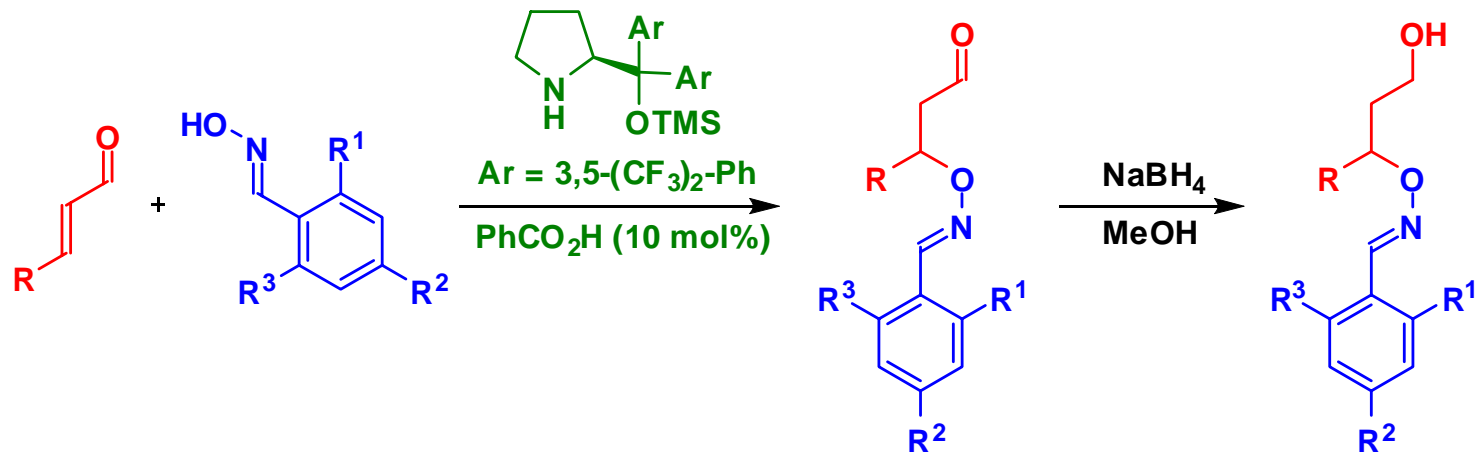
N-Heterocycle conjugate addition to α,β -unsaturated aldehydes



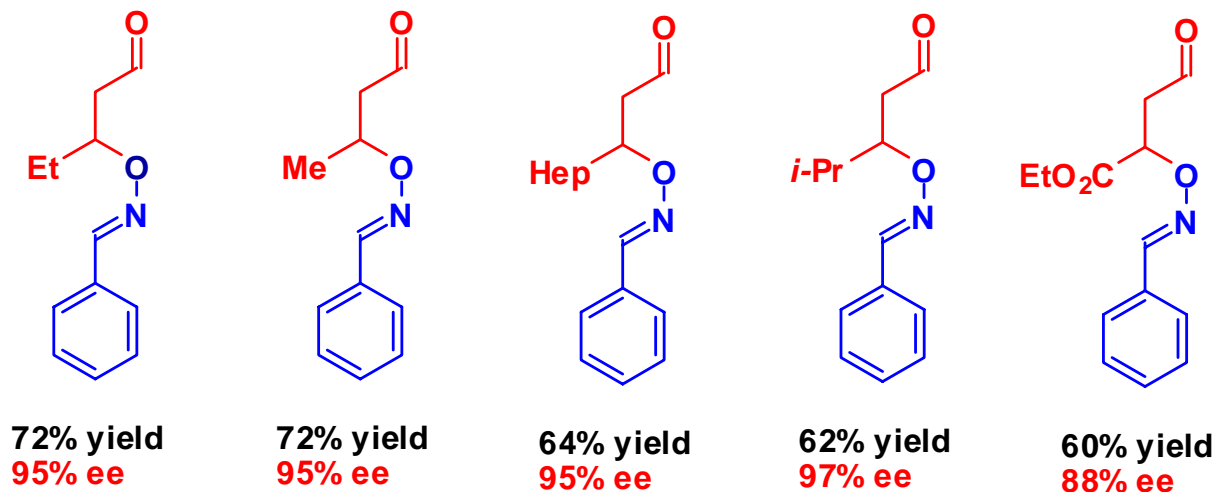
β -Hydroxylation of α,β -unsaturated aldehydes



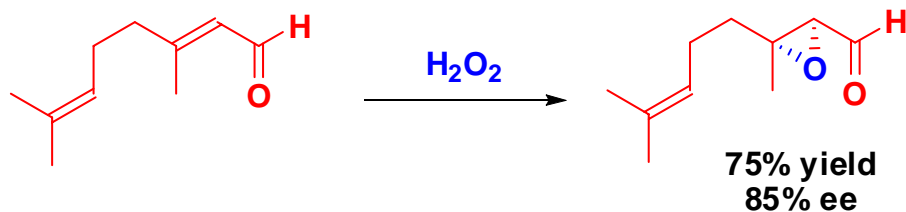
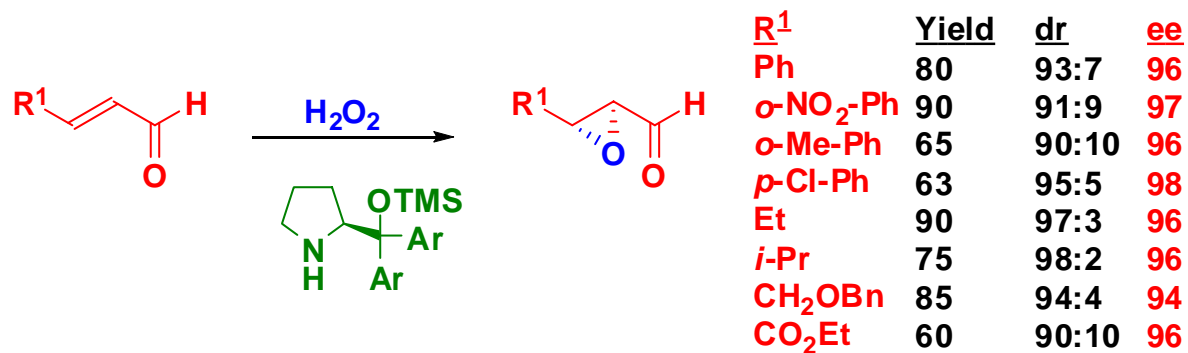
β -Hydroxylation of α,β -unsaturated aldehydes



anitnflammatory agents
penicillin and cephalosporin analogues
sex phermone analogues

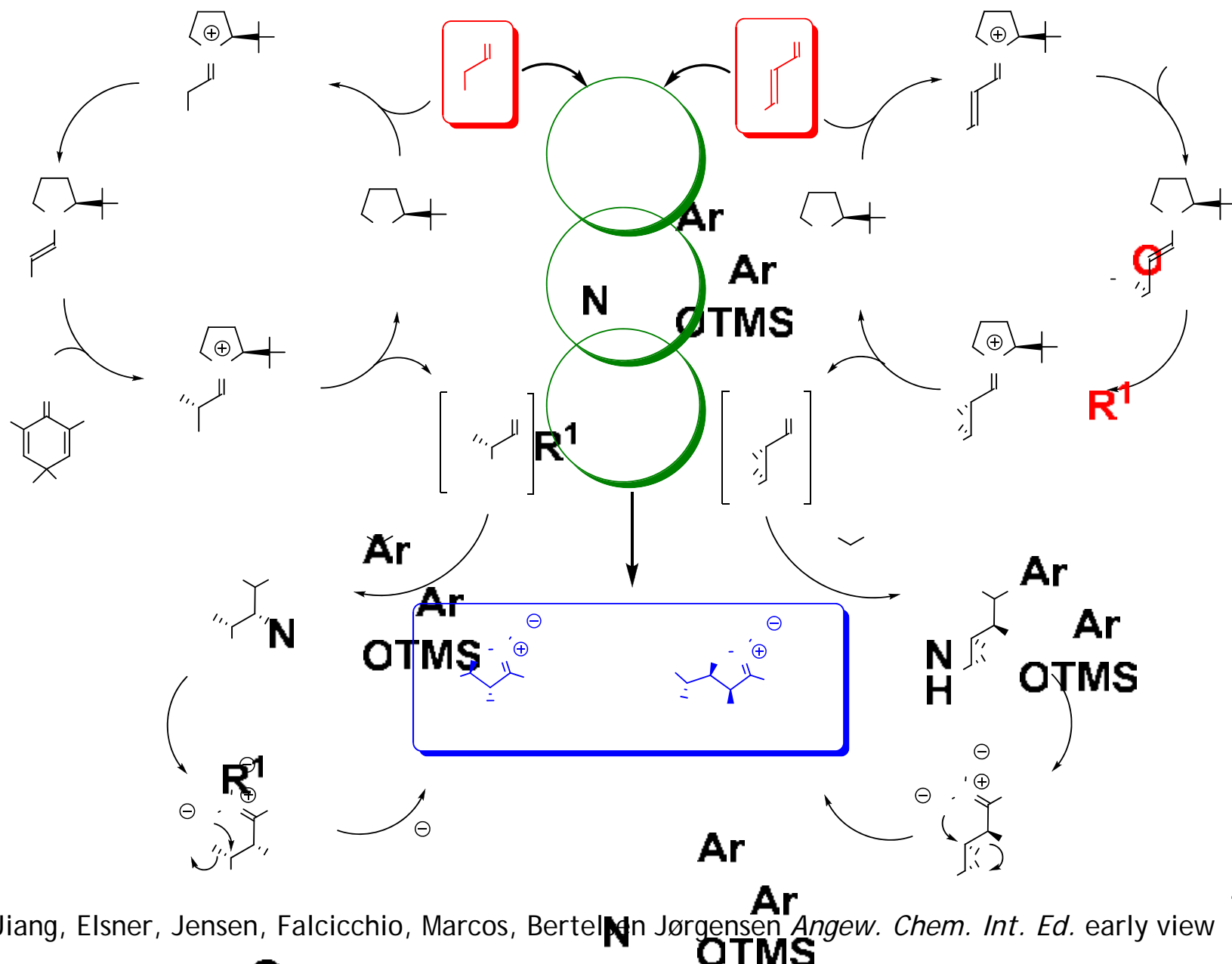


Asymmetric epoxidations

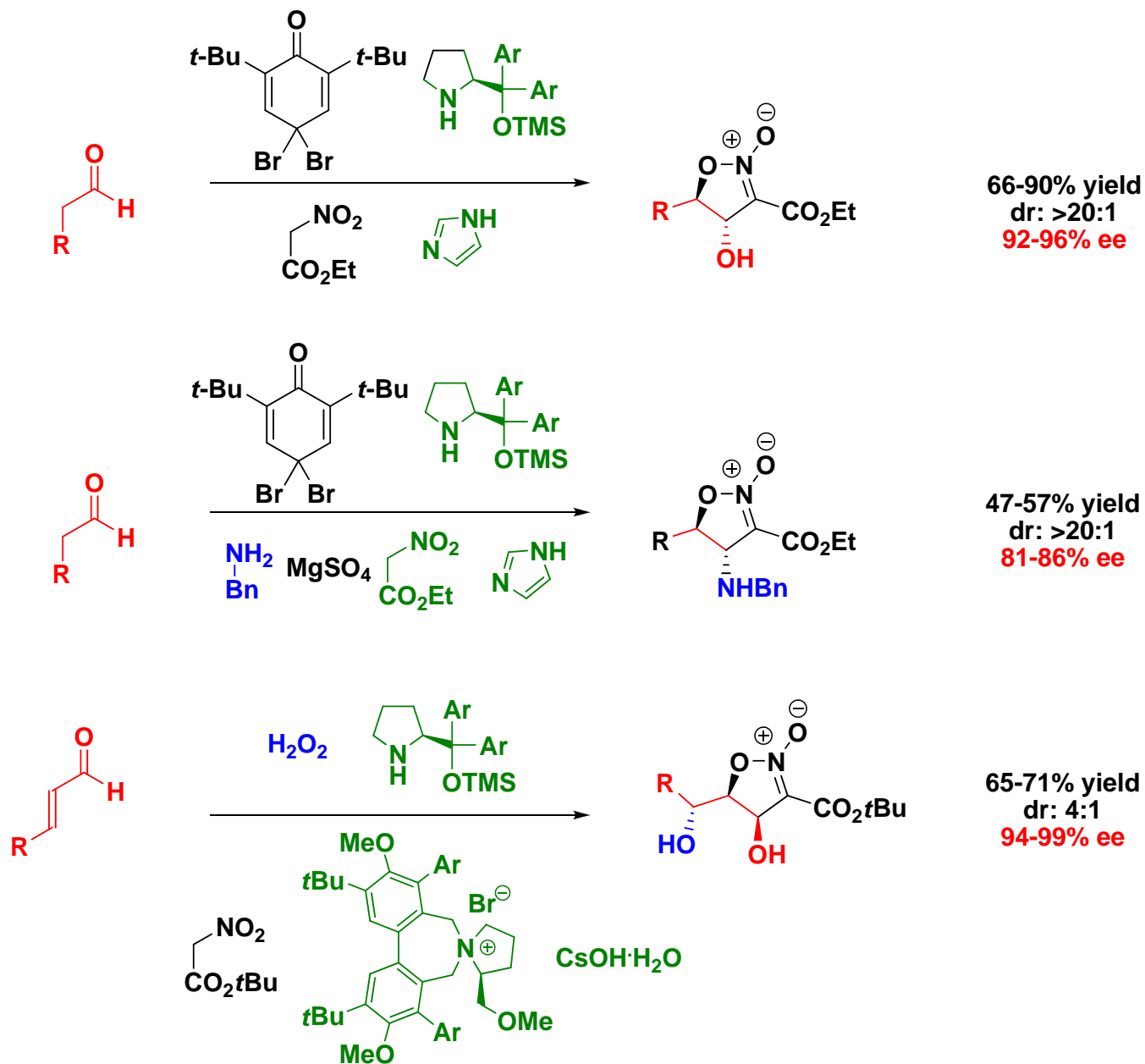


Marigo, Franzén, Poulsen, Zhuang, Jørgensen *J. Am. Chem. Soc.* 2005, 127, 6964

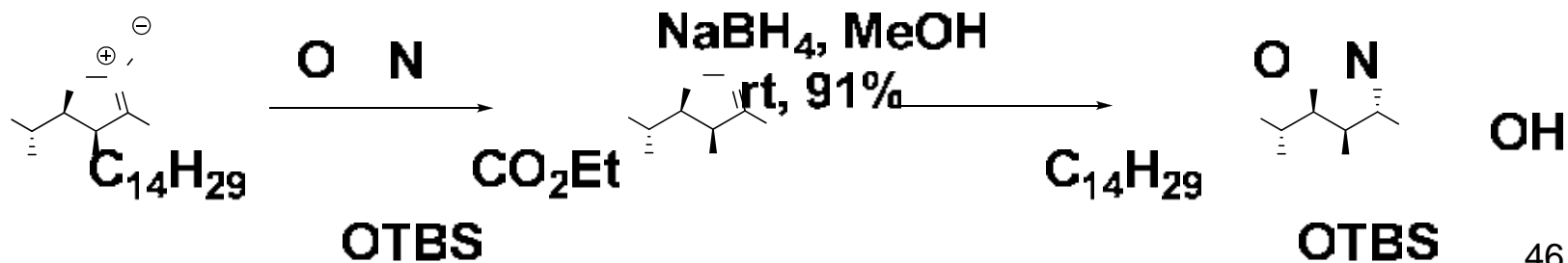
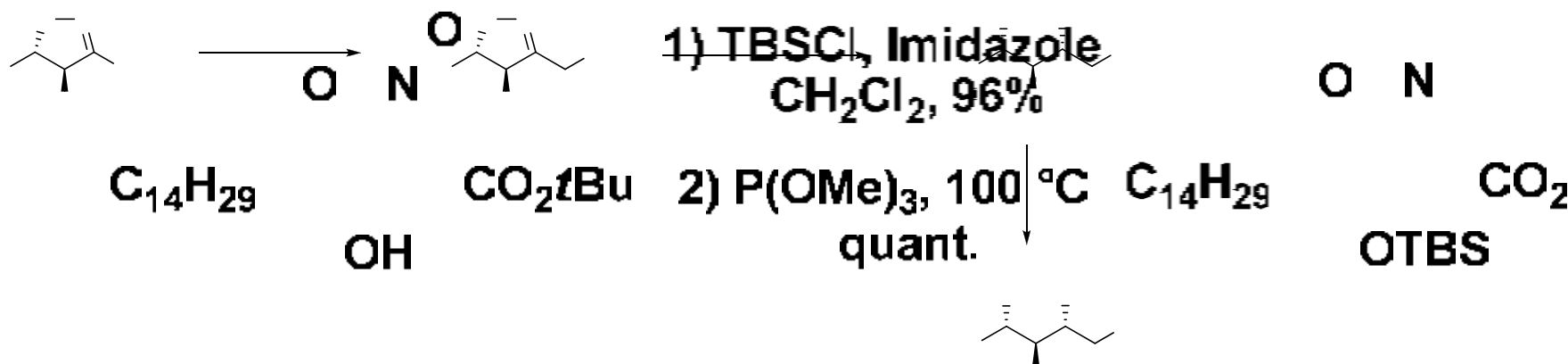
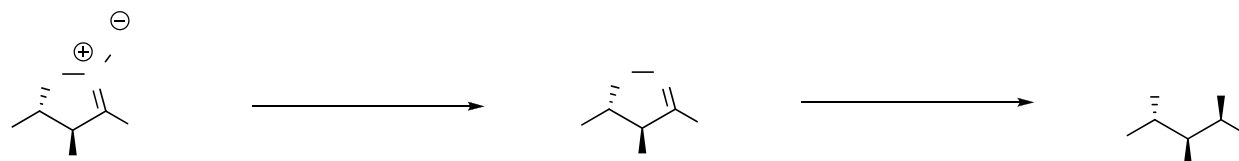
Organocatalytic leaving group strategy



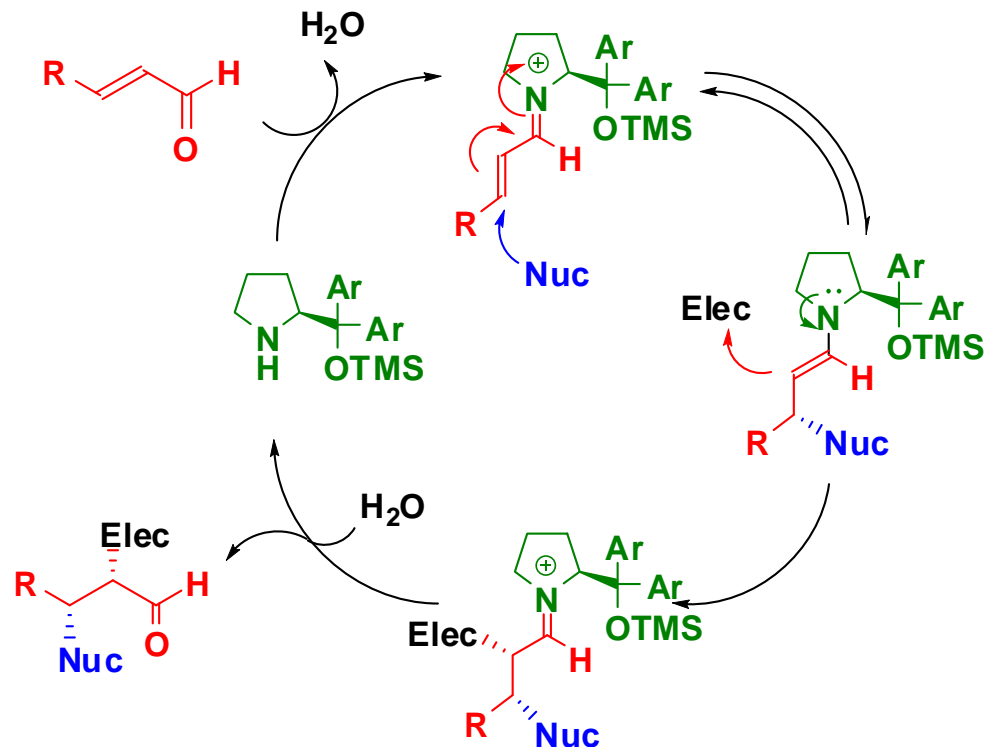
Scope



Potential

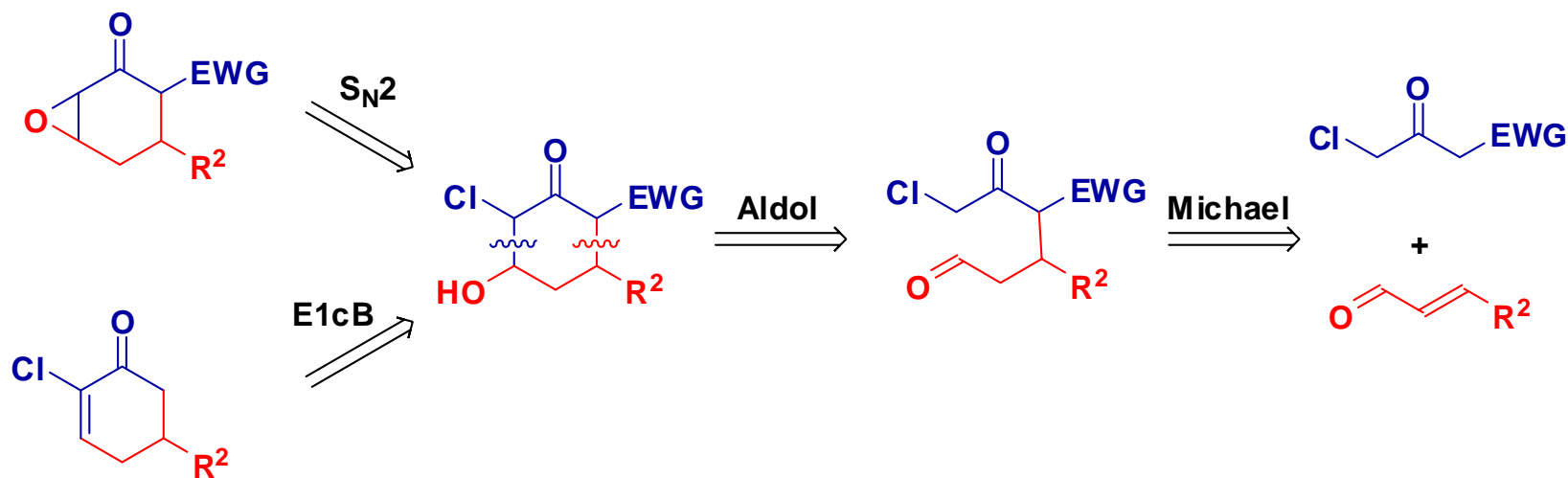


Organocatalyzed multicomponent reactions

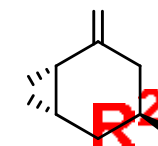
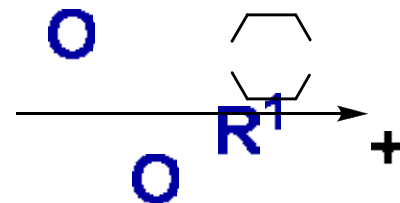
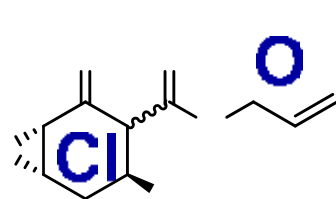
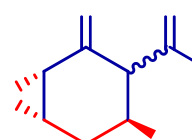
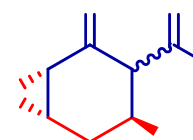
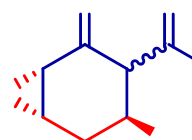
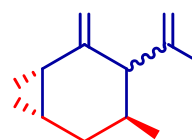
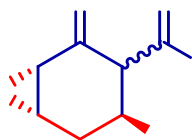
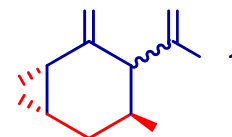
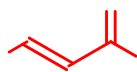
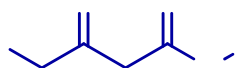
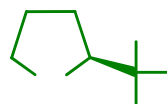


Yang, Hechavarria, List *J. Am. Chem. Soc.* 2005, 127, 15036
Huang, Walji, Larsen, MacMillan *J. Am. Chem. Soc.* 2005, 127, 15051
Marigo, Schulte, Franzén, Jørgensen *J. Am. Chem. Soc.* 2005, 127, 15710

Organocatalytic domino Michael-aldol reactions



Scope



O

H

O

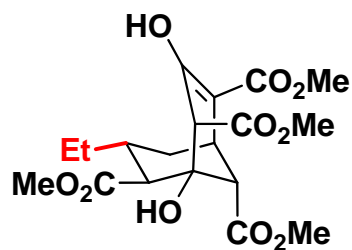
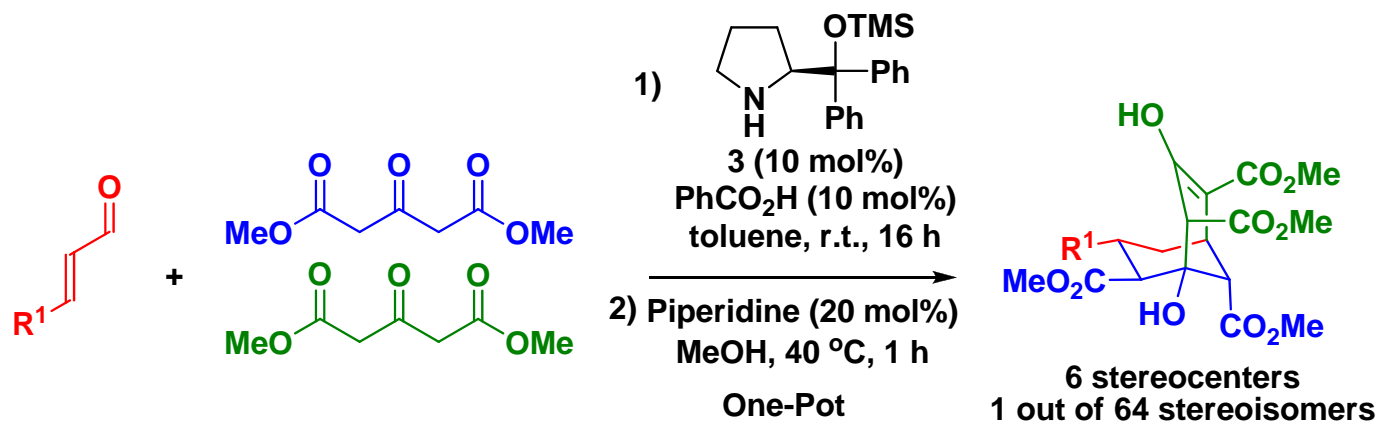
O

O

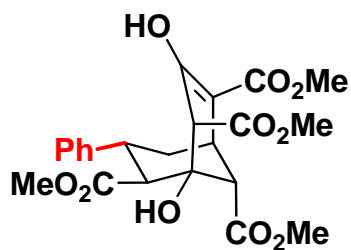
O

O

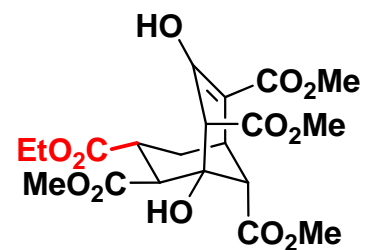
How to have six and 1 out of 64!



48% yield
dr >99:1
94% ee



70% yield
dr >99:1
94% ee



38% yield
dr >99:1
89% ee

Acknowledgements

Lukasz Albrecht, José Alman, Stephan Bachmann, Søren Bertelsen

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X-ray: Jacob Overgaard